

M. A. EL-SAYED

PROFESSIONAL DOSSIER

Undergraduate Education:

B Sc: Ein Shams University, Cairo, Egypt.

Graduate & Post-graduate Education:

Ph. D.: Florida State University, Tallahassee, Florida,
Research Associate, Yale University,
Research Associate, Florida State University,
Research Associate, Harvard University,
Research Associate, California Institute of Technology,

Positions Held:

International Committees membership:

A. University of California (LA):

Assistant Professor	1961-1994
Associate Professor	1964-1967
Professor	1967-1994

B. Georgia Institute Of Technology:

Juilius Brown Professor	1994- now
Regent Professor	2000-now

C. Visiting Professorial Positions:

American U. of Beirut,	1967-1968 (Visiting Professor)
University of Paris,	1968 (Visiting Scholar)
California Institute of Technology, Pasadena	1980 (Sherman Fairchild Distinguished, Scholar)
Technical University of Munich	1981 (Alexander von Humboldt Senior Fellow).

D. Editor-in-Chief: 1980-2004

The Journal of Physical Chemistry
American Chemical Society, Publication.

E. Editor: 1990-2000

International Review of Physical Chemistry.
Georgia Institute of Technology, 1994-
Georgia Institute of Technology, 2000 -

E. Miller Visiting Professor: 2007

University of California, Berkley, California, 2007

Recognitions: Election to Learned Societies and Membership of National and International Committees

Elected Member of the USA National Academy of Sciences, 1980.

Elected Associate Member of the Third World Academy of Sciences, 1984.

Selected member of the USA-Chinese Academy of Sciences Exchange Program, Summer 1984

Elected Member and Vice President of the Physical Chemistry Division of IUPAC, 1985-1989 and 1991-1995.

Elected Fellow of the American Academy of Arts and Sciences, 1986.

Elected Member at Large and Vice Chairman of the U.S. National Research Council Committee for the International Union of Pure and Applied Chemistry, 1987-1991.
Elected to the Board of Trustees of Associated Universities, Inc. (Managing Brookhaven National Laboratory and the National Radio Astronomy Observatory), 1988-1991.
Elected chairman of the U.S. National Research Council Committee for the International Union of Pure and Applied Chemistry, 1990-1992.
Member of the Division of Physical Chemistry, IUPAC, 1993-1995.
The Foreign Councilor of the Institute of Molecular Science, Okazaki, Japan, 1994-1997.
Member of the Board of Chemical Sciences, National Research Council, USA. National Academy of Sciences, 1994-1997.
Elected member of the American Association for the Advancement of Science (1999).
Elected to be ARegent Professor, Georgia Institute of Technology, 2000-
Elected Fellow of the American Association for the Advancement of Science, 2000.
Elected Fellow of the American Physical Society, 2000.
Honorary Fellow of the Chemical Research Society of India, 2000
Member of the Basic Energy Sciences Advisory Committee (BESAC) of the Dept. of Energy 2001-2011.
Member of the Mathematical and Physical Science Advisory Committee of NSF, 2003-2006
Elected Member of the U.S. Academy of Sciences, 1978
Elected Member of the American Physical Society, 1998
Member of the International Council of Scientific International Unions of Pure and Applied Chemistry
Member of ACS National Awards Committee for the Irving Langmuir Award in Chemical Physics 2005 –
Member of the External Advisory Board to the Michigan Nanotechnology Institute for Medicine and Biological Sciences 2006 –
Member of the Jean Sindab Endowment Advisory Board at Winship Cancer Institute, 2006
Inaugural Elected Fellow of the American Chemical Society, 2009 (First year this is given)
Elected Honorary Member of the Indian Chemical Society, 2008.
Elected Honorary Fellow of the Chinese Chemical Society, 2009.

Awards And Recognitions

The Distinguished Teaching Award, UCLA, 1964
The Alfred P. Sloan Fellowship, September 1965-September 1971
John Simon Guggenheim Memorial Foundation Fellowship, 1967-1968
Fresenius National Award in Pure and Applied Chemistry, April 1967
The McCoy Research Award, UCLA, October 1969
The American Chemical Society, California Section, Gold Medal Award, 1971
Sherman Fairchild Distinguished Scholar at Caltech, Fall Quarter 1980
Recipient of the Egyptian American Outstanding Achievement Award, 1988
Recipient of the 1990 King Faisal International Prize in the Sciences (Chemistry)
Recipient of the 1990 Tolman Award of the Southern California Section of the American Chemical Society
Selected member of the Advisory Committee for Chemistry Division of NSF, 1990
The 1991 UCLA Faculty Research Lecturer for 1990-1991 *Recipient of the Harris Award, University of Nebraska, Lincoln, NE 1995*
Recipient of the 1997 American Chemical Society Award, Tennessee Section, 1997
Elected member of the physical Society 1998.
Recipient of the 1999 American Chemical Society Award, Florida Section, 1999
Recipient of the Richard Medal of the Northeast Section of the American Chemical Society, 2000.
Recipient of the Langmuir Award in Chemical Physics of the American Chemical Society, 2001

The Georgia Tech 2007 Distinguished Professor of the year Award.
The 2007 USA National Medal of Science (Chemistry), 2008.
The Receptient of Honorary Doctors Degrees from: The Hebrew University, Israel (1980); the Mansoura Medical College, Egypt (2008); Alexandria University College of Medicine, Egypt (2008), Beny Swaif University, Benha University (2010) and Ain Shams University, 2013.
The Zewail prize in Moecular Sciences American Chemical Society (2009),
The Egyption Medal of the Republic of the First Class, 2009.
Selected to be among the top 100 most influential Persons in the State of Georgia, USA in 2011.
Selected to be among the 100 most influential Moslems in the state of Georgia, USA.in 2014.
On President Obama's USA National Medal of Science selection Committee (2014-2016).

Special Named Lectures

1. The Albert Noyes Visiting Lecture, University of Texas, Austin, Texas (1972)
- 2 The DuPont Lecture, Georgia Institute of Technology (1975)
- 3 Sierra Nevada Section Speaker, (American Chemical Society), University of Reno (1977)
- 4 The Industrial Users Lecture, California Institute of Technology (1978)
- 5 Bircher Lecture in Chemistry, Vanderbilt University (1980)
- 6 The Reilly Lecture, Notre Dame University (1981)
- 7 Distinguished Speaker Series, University of Utah (1982)
- 8 The Dreyfus Distinguished Visiting Scholar Lecture, Williams College (1982)
- 9 Joint Distinguished Lecture, Carnegie-Mellon University (1983)
- 10 Visiting Lecture Series, Columbia University (1983)
- 11 Mary E. Kapp Lecture, Virginia Commonwealth University (1984)
- 12 Arthur D. Little Lecture, Northeastern University, Boston (1984)
- 13 The Milton Kahn Lecture, University of New Mexico (1986)
- 14 The Randolph T. Major Lecture, University of Connecticut (1986)
- 15 The Kolthoff Lecture, University of Minnesota (1988)
- 16 Distinguished Summer Lecture Series, Northwestern University (1988)
- 17 Shell Chemistry Lecture, Florida State University, (1989)
- 18 The Chinese Academy of Sciences Distinguished Lecture Series (Taiwan) (1991)
- 19 The Edward Lee Lecture, University of California, Irvine (1991)
- 20 The 1991 Italian Academy of Science Speaker in Chemical Physics (1991)
- 21 The American Chemical Society Midwest Section/University of Wisconsin Physical Chemistry Colloquium. (1992)
- 22 Occidental Chemical Corporation Lecture, State University of New York, Buffalo (1992)
- 23 The Welch Foundation Lecture Tour (1993)

- 24 Charles A. Coulson Lecture Series, University of Georgia, Athens, GA (1994)
- 25 Advances in Chemical Physics Lecture, Rutgers, NJ (1995)
- 26 The Harris Lecture, University of Nebraska (1995)
- 27 The 1995 Boston College University lecture (1995)
- 28 Abbot Lecture, Northern Illinois University (1997)
- 29 Lind Lectures, Oak Ridge National Laboratory, Oak Ridge, TN (1997) and the University of Tennessee, Knoxville, TN (1997).
- 30 Frontier Lectures, Texas A&M (1997)
- 31 Annual Material Science Lecture, Material Sciences Dept, University of Pennsylvania (1997)
- 32 Nieuwland Lectures, Notre Dame University, South Bend, IN (1999)
- 33 Frontiers in Science Lecture, Florida Atlantic University, Boca Raton, FL (2000)
- 34 The Richard Medal Acceptance Speech, Harvard University, (2000)

- 35 Closs Lecture, University of Chicago (2000)
- 36 Weissberger/Williams Lecture Series, Eastman Kodak Co., Rochester, NY (2000)
- 37 The Foster Lecture, University of Buffalo (2000)
- 38 The H. Willard Davis Lecture, University of South Carolina (2000)
- 39 The Douglas G. Hill Memorial Lecture, Duke University (2001)
- 40 Lemieux Lecture, University of Ottawa, Canada (2002)
- 41 Eyring Lecture, Arizona State University (2004)
- 42 The Chancellor's Lecture, LSU (2005)
- 43 The Distinguished Lecture, Nanotechnology Center, UT, Austin (2005)
- 44 George B. Kistiakowsky Lecture, Harvard University, Cambridge, MA (2005)
- 45 E.U. Condon Lecture, Boulder, CO (2005)
- 46 2005 Frontiers in Nanotechnology Lecture, Northwestern University, Chicago, IL (2005)
- 47 The 2006 Pollak Lecture, The Technion, Haifa, Israel (2006)
- 48 The Zhong-Guan-Cun Forum on Condensed Matter Physics, Sept 17,2007 at the Institute of Physics, Chinese Academy of Science, Beijing, China.
- 49 The Wuhan Institute for Science and Technology Public Lecture, Sept 20, 2007, Wuhan, China
- 50 The Nanqiang Lecture, Xiamen University, Xiamen, China, Sept 26,2007
- 51 51The 2006 Pollack Lecture, The Technion, Haifa, Israel (2006)
- 52 The Zhong-Guan-Cun Forum on Condensed Matter Physics at the Institute of Physics, Chinese Academy of Science, Beijing, China (2007)
- 53 The Wuhan Institute for Science and Technology Public Lecture, Wuhan, China (2007)
- 54 The Nanqiang Lecture, Xiamen University, Xiamen, China (2007)
- 55 The Franklin Lecture, Rice University, Houston, TX (2008)
- 56 The 2012 G. N. Lewis Lecture, University of California, Berkeley (2012)
57. The Vanderbilt NanoCenter Chemistry Lecture (2013)
58. The Lord Spectroscopy Lecture MIT, (2014)

Regular Research Seminars at Research institutes and National meetings:

In the past 15 years, Professor El-Sayed gave one or two invited talks at Symposia at the two American Chemical Society annual Meetings.

Seminars Given Since Moving To Georgia Tech

- 2006 Valdosta State University; Valdosta, GA
 University of Georgia; Athens, GA
 University of Nebraska; Lincoln, NE
 University of California; Irvine, CA
 University of California; Los Angeles, CA
 The Hong Kong University of Science and Technology; Kowloon, Hong Kong
 Pierre-and-Marie-Curie University; Paris, France
 University of Cairo; Cairo, Egypt
 Wayne State University; Detroit, MI
 Duke University; Durham, NC
 Tel Aviv University; Tel Aviv, Israel
 University of Cairo; Cairo, Egypt

- 2007 Mount Holyoke College; Boston, MA
Northwestern University; Evanston, IL
Stanford University; Stanford, CA
University of California; Berkeley, CA
Virginia Commonwealth University; Richmond, VA
University of Utah; Salt Lake City, UT
University of California; Santa Cruz, CA
The State University of New York at Buffalo; Buffalo, NY
Xiamen, China
Suzhou University; Shanghai, China
University of Oslo; Oslo, Norway
- 2008 Cairo University; Cairo, Egypt
University of Houston; Houston, TX
University of California; Los Angeles, CA
University of Manchester; Manchester, England
Huazhong University of Science and Technology; Wuhan, China
Changsha University; Changsha, China
Institute of Chemistry, Chinese Academy of Sciences; Beijing, China
Xiamen, China
Paris, France
Oslo, Norway
Copenhagen, Denmark
- 2009 Dubai, United Arab Emirates
University of Rhode Island; Kingston, RI
University of Cairo; Cairo, Egypt
University of Pennsylvania; Philadelphia, PA
Florida State University; Tallahassee, FL
University of Wisconsin; Madison, WI
California Institute of Technology; Pasadena, CA
Nile University; Giza, Egypt
Egyptian Petroleum Research Institute; Cairo, Egypt
- 2010 Texas Tech University; Lubbock, TX
University of Texas; San Antonio, TX
Pennsylvania State University; State College, PA
University of Akron; Akron, OH

- University of Calgary; Calgary, Canada
Massachusetts Institute of Technology; Cambridge, MA
- 2011 Texas A&M University; College Station, TX
University of Buffalo; Buffalo, NY
American University of Beirut; Beirut, Lebanon
Case Western Reserve University; Cleveland, OH
University of Virginia; Charlottesville, VA
- 2012 American University of Cairo; Cairo, Egypt
Saudi Arabia: Taiba University,
Omelkorah University
Clemson University; Clemson, SC
University of Santiago de Compostela; Santiago, Spain
University of California; Berkeley, CA
Cairo, Egypt
- 2013 University of North Texas; Denton, TX
Stanford University; Stanford, CA
Kuwait University; Kuwait City, Kuwait
Columbia University; New York, NY
Northwestern University; Evanston, IL
McGill University; Montreal, Canada
University of Toronto; Toronto, Canada
University of Alabama; Tuscaloosa, AL
Purdue University; West Lafayette, IN
Kansas State University; Manhattan, KS
- 2014 Vanderbilt University: Nashville, TN
Ain Shams University, Cairo Egypt.
Georgia State University.
Massachusetts Institute of Technology; Cambridge, MA
University of Central Florida.
Plenary talk at the Faraday Discussions of the
Royal Society of England, Bristol, England.

**Summary of Invited and Plenary Talks at National and International Meetings and
Gordon Research Conferences And National Scientific Meetings :**

~450 TALKS (up 2014)

TALKS AT RESEARCH INSTITUTES

A. Special Named Lectures (see above)

B. Regular Seminars OVER 210 SEMINARS

VI. RESEARCH ACTIVITIES

Research at the University of California at Los Angeles

(1961-1994)

1. The Radiative and Dynamic Properties of the Lowest Triplet State of Molecules:

Research on the triplet state of molecules has been very active during the past twenty years and pursued by optical and magnetic resonance spectroscopists as well as by photochemists and photobiologists. The contribution of our group towards an understanding of the radiative, nonradiative, magnetic and structural properties of the triplet state has been extremely valuable and extensive. We have applied known techniques as well as developed new ones to obtain new fundamental information concerning the properties of molecules in the different spin levels of the lowest triplet state of prototype molecules.

From the results on the polarization of phosphorescence, our group was the first to obtain detailed interaction schemes responsible for the previously known heavy atom effects on the phosphorescence properties. One of our papers was one of two that came out simultaneously introducing the method of "magnetophotoselection," now commonly used in triplet ESR research. In 1968, we began to develop the field of phosphorescence microwave double resonance. We showed how to use these techniques to determine the radiative, nonradiative, magnetic and structural properties of molecules in the different spin levels of the lowest triplet state. Most notably is our work in which we used the spin label of triplet molecules to determine the mechanism of the nonradiative electronic energy relaxation process from the lowest excited singlet state to the lowest triplet state. We showed that the effect of small magnetic field on the net spin polarization of the system at low temperature enables the determination of the mechanism of this important nonradiative process. We then applied this new valuable technique to elucidate the mechanism in a number of important systems. In N-heterocyclic systems, the results are found to be consistent with the selection rules we proposed over ten years earlier (The El-Sayed's Rule:

Wikipedia, the free encyclopedia) Using Stark effect on the double resonance signal, we were the first to measure and account for the difference in the dipole moment of some polar molecules in the different spin levels of the lowest triplet state. Using double resonance methods, we were the first to show that in some solid-state photochemistry, the reactivity of molecules in different spin levels of the lowest triplet state can be different.

2. Lasers and Molecular Dynamics:

During the past ten years, our group has been successful in using the different properties of lasers to develop new techniques and to study intra and inter-molecular dynamics in different systems, ranging from isolated gas molecules to the photosynthetic system of bacteriorhodopsin. Below a summary of the important contributions made during the past few years is given:

a.Multiphoton Ionization and Dissociation Spectroscopy:

Our group was among the first two groups to use two-color lasers and ionization detection to measure the lifetime of excited states and the ionization potential of molecules. In two-photon-multiphoton ionization spectroscopy, our group was the first to use polarization techniques in assigning two photon transitions of large polyatomic molecules. This technique was then applied to identify the type of the electron (ρ vs. nonbonding) having the lowest ionization potential in heterocyclic compounds.

In laser multiphoton ionization-dissociation, a technique was developed to differentiate between mechanisms involving ionization followed by dissociation and those involving dissociation followed by ionization. In this technique two-color lasers are used and the time-of-flight mass spectrum is studied as a function of the repelling voltage applied in the ionization-extraction zone.

Recently, by combining two-color picosecond laser techniques with mass spectrometry, our group was able to determine the rates of energy redistribution prior to ionic fragmentation in mass spectrometry. This technique makes it potentially more profitable to study energy redistribution in ions than in neutral species. In addition, it offers an experimental method to test the applicability of the quasi-equilibrium theory proposed in 1944 to describe the observed mass spectra of molecules.

b.Switching of Energy Transfer Mechanism with Distance in Disordered Solids:

Using time-resolved laser luminescence line narrowing (LLN) techniques, the mechanisms of energy transfer between the same chemical species in different sites have been investigated in disordered solids. In pure solids with large inhomogeneous linewidth, it was shown that by changing the wavelength of the existing laser, the average donor-acceptor distance can be changed continuously if the energy transfer studies are carried out at 1.6 K. Using this technique, a careful distance dependence of the energy transfer mechanism in some of these solids at low temperature has been investigated. Our group was the first to demonstrate switching in an energy transfer mechanism with distance. Thus while triplet-triplet energy transfer occurs via an electron-exchange mechanism at short distances for 1-chloro-4-bromonaphthalene solid at 1.6 K, a dipole-dipole mechanism dominates at distances larger than ~ 10 .

3,Time-Resolved Raman Spectroscopy:

Our group has developed a number of techniques that enabled the determination of the resonance Raman spectra of transients formed in the milli-, micro-, nano-, and picosecond time domains. In each time regime, a different type of lasers (c.w. or pulsed) or a different technique of sample excitation (static vs. sample flow technique) was used. In the different techniques, the time resolution is limited by either the laser pulse width or by the focal volume and the sample flow rates. Using the microbeam flow technique with picosecond lasers, our group was the first to report the resonance Raman spectrum of a picosecond transient.

The above techniques were developed in order to answer important questions about the mechanism of some photobiological changes. Bacteriorhodopsin, the second photosynthetic system in nature besides chlorophyll, was one of two systems studied with the time-resolved

resonance Raman technique that we have developed. We were the first to show clearly that retinal conformational changes indeed occur in a time shorter than 50 ps. We have also shown that, unlike chlorophyll, intermolecular energy transfer does not take place in bacteriorhodopsin prior to photochemical changes, i.e., the absorption and the primary photochemical reaction (configuration) changes occur on the same molecule. In combination with optical time-resolved studies, the nature of the coupling between the retinal and the protein is examined.

4. Rate of Energy Redistribution in the Transition State

We have developed a simple technique using nanosecond pulsed lasers and a home built time of flight mass spectrometer to determine the rate of energy redistribution in molecules falling apart on the subpico-to-picosecond time scale. This is also the time scale for molecular rotation of some molecules. For these systems, an excited ensemble will have a distribution of lifetimes. The ones that dissociate immediately will have no time to redistribute any excess energy above bond dissociation into internal energy. The fragments will be anisotropically distributed and should have large recoil translation energy. Molecules that have time to rotate, before dissociation, will give fragments in different direction than those produced from rapid dissociation and will also have less recoil energies (as some of the excess energy is redistributed in internal energy during the time it takes the molecule to rotate). Thus by measuring the recoil energy of fragments detected along the laser polarization direction and perpendicular to it, the amount of energy that redistributes in a quarter of a rotation can be determined. In the actual experiment, the polarization ratio of the signal intensity of the fragments at different recoil velocity is determined and is related to the time dependent correlation function, from which the time dependence of the recoil velocity (and thus of the recoil translation energy) is determined. From the slope, the rate of energy redistribution is determined.

The dependence of the rate of energy redistribution on the excited state structure of different molecules is examined. For smaller molecules (e.g. ICN, HgI₂.) the dependence of this rate on the calculated potential energy surfaces and the effect of curve crossing is being studied.

5. The Structure and Stability of Mixed Gaseous Clusters

Metallic-Ionic Mixed Clusters:

The evaporation dynamics of (Ag_x)⁺ clusters in which different number of Ag atoms are changed into AgI molecules are examined. Questions like how does the binding of an AgI polar molecule to the Ag⁺(_{x-1}) cluster depends on its polarizability is to be answered. In addition, in clusters with more than one AgI molecule, do the polar molecules stay together or spread around the metallic part of the cluster? Do we get phase separation. What fraction of the Ag atom needs to be converted into AgI before phase separation between the metallic and the ionic part of the cluster occurs? Both theoretical and experimental studies are being carried on in order to understand this interesting system.

Mixed Ionic Clusters:

How much a cluster like [Cs(CsI)₁₄]⁺ is destabilized by introducing defects created by substituting one or more of the CsI molecules with another alkali halide molecules of different size (e.g. in [Cs(CsI)₁₃ ⋄ RbI]⁺? By studying the evaporation dynamics of the mixed cluster one can determine its relative rates of evaporation to give CsI or RbI from the relative mass peak

intensity of the corresponding daughter ions. By assuming the absence of barriers for the condensation process (which involves ion-dipole interaction), one can deduce from the kinetic data thermodynamic quantities from which the effect of impurity (RbI) substitution on the stability of the parent cluster can be determined. Furthermore, the energy difference between different "isomers" (resulting from the different sites that the RbI molecule(s) can occupy) can be determined. Using a modified polarizable electrostatic model of Martin et al., one can calculate the internal energies of these mixed clusters, as is done for pure clusters. A careful test of the model can be accomplished by calculating these small energy differences and comparing them to experiments.

Coulombic Explosion in Cluster Dications

Using Mass spectrometric techniques, the stability of dication metallic clusters is examined. The critical cluster size below which the cluster is unstable towards Coulombic explosion will be determined for different metal dication clusters. The effect of adding different atoms, e.g., halogens, on the stability of present clusters will be examined. Theoretical predictions of the critical size will be tested and more understanding of the cluster structures and the mechanism of their explosion will be accomplished.

6.The Molecular Mechanisms of the Proton Pump in the Photosynthesis of Bacteriorhodopsin (bR)

The Protein Catalysis of the Retinal Photoisomerization

Bacteriorhodopsin, a retinal protein membrane present in halobacterium halobium, is the other photosynthetic system in nature besides chlorophyll. Upon the absorption of light, it rapidly isomerizes leading to charge separation. The system then goes through a photocycle with a minimum of six intermediates before it returns to its original state. As a result, protons are pumped from inside the cell to the membrane surface, thus converting solar to electric energy, which is used in ATP synthesis.

We, and others have shown that the primary process in bR photocycle is indeed a retinal photoisomerization. It occurs extremely rapidly on the 0.5 ps time scale and only around the C₁₃ - C₁₄ bond. In solution, this process occurs on the nanosecond time scale and is not so highly specific. The question immediately arises regarding the molecular mechanism(s) by which the protein catalyzes this process and makes it so highly specific in bR (as well in rhodopsin).

In order to answer these questions, we are involved in two types of studies. In one, the rate of retinal isomerization of a number of mutants in which individual charged or hydrogen bonding amino acids are genetically replaced by nonhydrogen bonding neutral ones. This work is done in collaboration with Professor J. Lanyi of U.C. Irvine who prepares the different mutants.

In the second type of studies, the somerization of bound modified retinals (synthesized by Professor R. Liu of the University of Hawaii) are measured. The aim of these studies is to separate the effect of electrostatic interactions from the indirect sterric effects within the cavity as a factor in the catalyses and the specificity of the protein to the photoisomerization process.

Metal Cation Role in the Proton Pump Function of bR

The removal of metal cations from bR inhibits the proton pumping function. The question thus arises as to their role in making the molecular pump work.

A detailed study of the binding sites of the metal cations is underway. The importance of negatively charged amino acids (aspartates and glutamates) and hydrogen bonding residues to the binding is examined by a systematic genetic replacement of each of the important ones and examining the effect on the value of the binding constants of the different sites.

After the binding characteristic is understood, the effect of metal cation on the dynamical structural properties will begin. In these studies, an examination of the effect of binding of metal cations on the FTIR spectrum of bacterio-opsin and its mutants will be examined. In this manner, the amino acids most directly affected by the binding could be identified. A detailed examination of the effect of different metal cations on the energy and entropy of activation of both the deprotonation process as well as the proton transport process to the surface will be carried out. The latter will be extracted from the temperature dependence of the change of color of pH sensitive dyes attached to an amino acid at the opening of the proton channel.

A final determination of the location of metal cations in bR could be made by use of anomalous x-ray scattering in Brookhaven National Lab. In this work, metal cations whose K or L resonances are in the x-ray wavelength range taken of BNL synchrotron facility will be used. By subtracting the diffraction pattern obtained off from that on resonance only the scattering from the metal cations will be determined. If we do this experiment as one gradually fills in the different sites, it is hoped that the location of the different binding sites can be determined from the two dimensional projection map obtained from these studies. The results of the FTIR studies should be helpful in this complex effort.

B. RESEARCH CARRIED OUT AT THE GEORGIA INSTITUTE OF TECHNOLOGY, (1994-

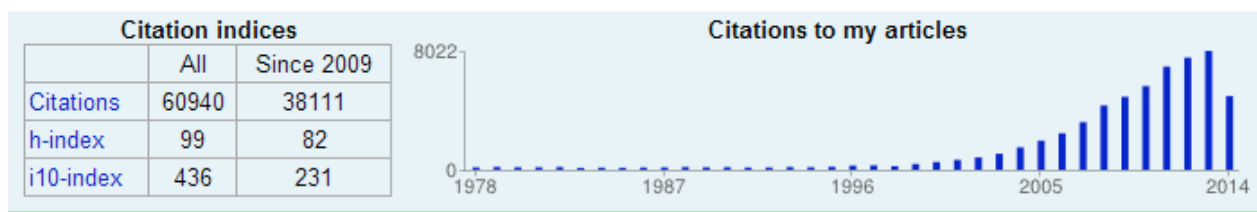
The following research projects are carried out in the Laser Dynamics Laboratory (LDL) directed by Professor El-Sayed (<http://ldl.gatech.edu>). LDL houses the most recent lasers and laser spectroscopic equipment for time resolved optical, Raman and IR studies. It also houses biochemical labs equipped with cell culture facilities, modern facilities of plasmonic imaging in time the single cells and their contents using either SERS spectra or Rayleigh images during the cell full life (between birth to division) or as it dies. Available for LDL researchers the use of the Georgia Tech clean room facilities with its High resolution TEM, electron-beam facilities for the synthesis of nanoparticles of different shapes, and its high resolution structural determination facilities.

The present research interests are focused on the synthesis, the determination of the structure and the properties of materials (semiconductor and metallic nanoparticles) of sizes confined to the nanoscale (between one and 100 nm). Applications in different technical fields ranging from nanocatalysis, nanosensors to cancer nano medicine and molecular cell-biology were developed and used.

Contributions to Nanoscience and Nanotechnology:

INTRODUCTION

Over the past twenty years, El-Sayed and his group have made unparalleled contributions to the fields of nanoscience and nanotechnology. Figure 1 below (taken in July of 2014 from Google Scholar citation): <http://scholar.google.com/citations?user=BMpDLscAAAAJ&hl=en>) shows the dramatic change in the annual number of citations with time of the published papers by his group before and after changing his research activity to the nanoscience field (in the mid-nineties). It also shows the large change in his h-index in the last five years, when the impact of focusing his nano-science research on nano-medicine and cancer research (the developing different techniques using gold and silver nanoparticles in the diagnosis, therapy and studying on single cell of cancer).



after getting into the nanoscience research in the late nineties until mid of 2014 (Scholar.google.com/citations).

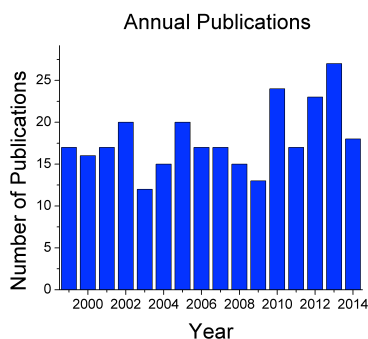


Fig 2. El-Sayed's group annual publications in the nanoscience research field until mid July of 2014.

He and his group have published more than 670 articles total by the mid of 2014 with half of this appeared in the last twenty years (The number of citations per paper published in the past few years has been in the hundreds (See the large exponential increase in the citations to his research papers in this period in fig 1 and the number of published papers in fig 2).

LDL research laboratory has been consistently ranked among the top chemical research laboratories in the world during the last decade (#4 among Academic laboratories according to the Times Higher Education and #17, among Academic, national, international and industrial laboratories as concluded by Thomson-Reuters: Refer to his website: <http://ldl.gatech.edu>). Figure 2 also shows the large

increase in the # of his annual published papers reached 27 papers in 2013 and extrapolated to over 30 papers in 2014. It is this excellent scientific creative performance as well as the research field that El-Sayed selected to do his research that were responsible for crowning him with the **highest national honor** a scientist in the USA can receive, this is the USA National Medal of Science (in chemistry) :nanoscience: nanocatalysis, and nanomedicine (please go to: <http://ldl.gatech.edu>.)

Early studies in LDL in this field included the unique optical and photo-thermal properties of gold and silver nanoparticles of different and discussed the potential use could be used to more easily Nano-science (e.g. in 1,2,3,4,5,6) that lead to the rapid development of this important field and some of its potential applications in the field of nanotechnology, e.g. in catalysis, in sensors and in the important field of medicine. This is the reason for the very high citations to his work. During the past five years alone, LDL publications have received 37,880 citations. : <http://scholar.google.com/citations?user=BMpDLscAAAAJ&hl=en>

On Jan. 29, 2010 - the Laser Dynamics Laboratory (LDL) (see <http://ldl.gatech.edu/>), whose director is Professor El-Sayed, was ranked #4 Worldwide among Academic chemistry research labs of the past decade by the Times Higher Education. LDL publications garnered an impressive 75.74 citations per paper over this period. On March 21, 2011, the Laser Dynamics Laboratory was ranked #17 worldwide among all chemistry research labs (Academic, National as well as the Industrial labs) during the past decade (2000-2010) by the Thomson Reuters. During this period. LDL published more highly cited manuscripts than any others in the top 20.

Below we will describe some of the LDL to nanoscience and its important impact on technology. Three fields this group has studied and observed new nano-scientific properties with some are shown to have important impact on the field of Nanotechnology.

References:

1. C Burda, X Chen, R Narayanan, MA El-Sayed, Chemistry and properties of nanocrystals of different shapes, Chemical reviews 105 (4), 1025-1102, 2005 (invited, 4400 citations)
2. S Link, MA El-Sayed, Spectral properties and relaxation dynamics of surface plasmon electronic oscillations in gold and silver nanodots and nanorods,, The Journal of Physical Chemistry B 103 (40), 8410-8426 (1999) (Invited, 2507 citations)
3. MA El-Sayed, Some interesting properties of metals confined in time and nanometer space of different shapes, Accounts of Chemical Research 34 (4), 257-264, 2001 INVITED (2080 citations).
4. .S Link, MA El-Sayed, Shape and size dependence of radiative, non-radiative and

photothermal properties of gold nanocrystals, International Reviews in Physical Chemistry 19 (3), 409-453 (Invited, 1330 citations)

5. Shape-controlled synthesis of colloidal platinum nanoparticles, TS Ahmadi, ZL Wang, TC Green, A Henglein, MA El-Sayed, SCIENCE-1924-1925, 1996 (2013 citations).

6. P. K. Jain, X. H. Huang, I. H. El-Sayed and M. A. El-Sayed. (2008) Noble Metals on the Nanoscale: Optical and Photothermal Properties and Some Applications in Imaging, Sensing, Biology, and Medicine. Accounts of Chemical Research 41, 1578-1586. (Invited, 1221 citations)

7. IH El-Sayed, X Huang, MA El-Sayed, Surface plasmon resonance scattering and absorption of anti-EGFR antibody conjugated gold nanoparticles in cancer diagnostics: applications in oral cancer, Nano letters (5), 829-834, (1120 citations, The most cited paper in the Journal of the American Chemical Society in the whole year)

We shall give short account of the contributions of El-Sayed and his group to the different studies and applications they made to the three different fields of the nanoscience; nanocatalysis, nanosensors and nanomedicine.

1.NanoScience: The new Properties of Material Confined in Time and Space of Different Shapes:

The type of electronic motion in matter determines its property and thus its uses in our everyday life. This motion itself is determined by the forces acting on the electrons and thus the space in which they are allowed to move. The difference between a metal, a semiconductor and an insulator lies in the fact that the electronic motion is highly delocalized, slightly confined, and highly confined, respectively. One thus expects that if we reduce the size of material to below its naturally allowed characteristic length scale new properties should be observed which are different from that of the macroscopic material as well as of their building blocks (atoms or molecules). This size is on the nanometer length scale.

a.Ultrafast Electron-Hole Dynamics in Semiconductor Nanoparticles

The type of electronic motion in matter determines its property and thus its uses in our everyday life. This motion itself is determined by the forces acting on the electrons and thus the space in which they are allowed to move. The difference between a metal, a semiconductor and an insulator lies in the fact that the electronic motion is highly delocalized, slightly confined, and highly confined, respectively. One thus expects that if we reduce the size of material to below its naturally allowed characteristic length scale new properties should be observed which are different from that of the macroscopic material as well as of their building blocks (atoms or molecules). This size is on the nanometer length scale.

b.Shape Controlled Synthesis, Stability and Self Assembly of Metallic

Nanoparticles

Different methods are developed to synthesize, reshape and study the self-assembly characteristics of gold, silver and transition metal nanoparticles. High resolution TEM is used to follow the shape distribution and their dependence on the preparation condition and to study their thermal stability. From the results, the mechanism of shape controlled growth and self-assembly are elucidated.

c, Photothermal Stability of Metallic Nanoparticles

Metallic nanoparticles of non-spherical shape have been found to undergo a sharp transformation into the more thermodynamically stable spherical shape induced by pulsed laser excitation. A study is directed to understand mechanism and relevant time scales involved in this photothermal shape transformation. By adjusting the power and wavelength of the pulsed laser, the size and shape distribution of non-spherical nanoparticles can be changed and the nanoparticles can be reshaped.

d. Optical and Nonradiative Properties of Assembled Metallic Nanoparticles

Three methods are used for the assembly of metallic nanoparticles: 1) From colloidal solution of highly mono-dispersed sample, 2) Nanosphere lithography, and 3) Electron beam Lithography. Metallic nanoparticles assembled in monolayer periodic arrays present opportunities to study both unique properties of individual nanoparticle and collective properties of coupled nanoparticles. The current fundamental research in the Laser Dynamics Laboratory on metallic nanoparticle arrays includes: 1) Ultrafast dynamics of coherent vibration induced by femtosecond laser; 2) Effects of electronic coupling between nanoparticles on the optical and electronic properties; 3) Medium effects on electron-phonon and phonon-phonon dynamics. 4) The laser photothermal heating of gold and silver nanoparticles can result in the heating or melting of the medium surrounding the nanoparticle. Alternately, the nanoparticles may themselves melt or atomic ablation may take place. We are studying the dependence of the nature of photothermal heating of gold and silver nanoparticles on the rate of heat deposition by varying the photothermal laser pulse energy and the pulse duration as well as the effect of cw laser on photo-thermal destruction of cancer cells.

The current fundamental research in the Laser Dynamics Laboratory on metallic nanoparticle arrays includes:

Ultrafast dynamics of coherent vibration induced by femtosecond laser
Effects of electronic coupling between nanoparticles on the optical and electronic properties.

Medium effects on electron-phonon and phonon-phonon dynamics.

The laser photothermal heating of gold and silver nanoparticles can result in the heating nanoparticles may themselves melt or atomic ablation may take place. We are studying the dependence of the nature of photothermal heating of gold and silver nanoparticles on the rate of heat deposition by varying the photothermal laser pulse energy and the pulse duration on biological cell for

possible cancer cell treatment.

2. NanoTechnology: Potential Applications of Nanoparticles

a. Applications in Nanocatalysis:

Nanoparticles are potentially attractive catalysts since they have a large surface-to-volume ratio and high surface energy compared to bulk catalytic materials. In addition, metal nanoparticles of different shapes have different crystallographic facets and different fraction of atoms located on their corners and edges, which makes it interesting to study the effect of nanoparticle shape on the catalytic activity of various reactions. It is also important to note that having very active surface atoms could make the nanoparticles unstable during the course of its catalytic function. In the case of the early stages of the electron transfer reaction, we found that tetrahedral platinum nanoparticles are the most catalytically active and have the greatest fraction of surface atoms on their corners and edges, while the cubic platinum nanoparticles are the least catalytically active and have the lowest fraction of surface atoms on their corners and edges. During the full course of the reaction (2 days), it is observed that distortions in the corners and edges of both the tetrahedral and cubic platinum nanoparticles take place. In addition, the rate of dissolution of corner and edge atoms is found to be faster for the tetrahedral nanoparticles. Shape dependence of the catalytic rate on the fraction of atom on corners and edges observed at the early stage of catalytic electron transition reaction Radha Narayanan and Mostafa A. El-Sayed, *Nano Letters*, 2004, 4(7), 1343-1348; Radha Narayanan and Mostafa A. El-Sayed, *Journal of the American Chemical Society*, 126(23), 7194-7195 (2004). As the catalytic reaction continues in colloidal solution, shape changes take place eliminating sharp edges and corners (to eventually make spherical shape). Radha Narayanan and Mostafa A. El-Sayed, *Journal of Physical Chemistry B*, 2004, 108(18), 5726-5733; Radha Narayanan and Mostafa A. El-Sayed, *Journal of the American Chemical Society*, 125(27), 8340-8347 (2005).

heating or melting of the medium surrounding the nanoparticle.

The sharp corners and edges are valency unsatisfied (surrounded with lower number of bonded atoms compared with the atoms on the faces) which increases their chemical activity. Nanocatalysts of different shapes with sharp corners and tips have been prepared and examined in catalysis such as cubes², tetrahedral², and multi-armed stars⁶. Although such unique design introduced an exciting catalytic

Noble metallic nanoparticles such as palladium and platinum are widely used in catalysis, due to their high catalytic efficiency for many chemical and electrochemical reactions.¹ Different shapes of the metallic nanocatalysts have been prepared with different designs.^(2,4) The main goal of controlling the design of the nanocatalysts is to increase their catalytic efficiency, which reduces the production cost; one of these designs is to introduce sharp corners and edges in the shapes of the nanocatalysts.⁵ However, the atoms in the sharp corners and edges are valency unsatisfied (surrounded with lower number of bonded atoms compared with the atoms on the faces) which increases their chemical activity⁽⁶⁾. Nanocatalysts of different shapes with sharp corners

and tips have been prepared and examined in catalysis such as cubes², tetrahedral², and multi-armed stars⁶. Although such unique design introduced an exciting catalytic efficiency for such shapes, possible rounding of the corners and the edges during the catalysis is a serious challenge that has not been resolved. In order to resolve the reshaping challenge in using sharp corners and edges nanocatalysts, we have designed the nanocatalysts to be of hollow structure⁹⁻¹¹. Nanocatalysts of hollow design presented an excellent catalytic efficiency which based on the confinement of the reactive species either intermediates or reactants inside the hollow nanocatalysts during the catalysis, this confinement increases the collision probability with the surface of catalyst and enhances the efficiency of the catalysis reaction⁹. Hollow nanostructures of different designs have been prepared of single shell and showed high catalytic efficiency such as gold⁹, platinum³, palladium⁴ and double shells³ such as platinum-palladium, gold-palladium⁹, and gold-platinum¹⁰.

In order to understand the relationship between the photo-catalytic efficiency of a semiconductor and electron-hole charge separation on the nanoscale, we have used the femtosecond pump-probe technique to study the excited electronic state decay dynamics in Cu₂O nano-shell coating gold nanoframes¹¹. The lifetime of the electron decay of Cu₂O of different thickness around the gold nanoframes were compared with their photocatalytic efficiency¹².

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B. Applications in Nanophotonics

When photons interact with nanoscale materials (semiconductor and metal), many new physical phenomena will be observed which are not present in corresponding bulk materials. Nanophotonics provides opportunities for making ultra-small optoelectronic devices which have great performance in lasing, sensing, and communication. In our group, we have made two-dimensional metallic nanoprism arrays with well-defined size, shape, and interparticle separation using nanosphere lithography technique. These samples could be used in surface-enhanced spectroscopy, plasmonic devices, and sensors due to its tunable optical properties. We have observed modulation of the color of plasmonic nanoparticles with coherent lattice vibration due to the dependence of the surface plasmon frequency on lattice volume change with the modulation period proportional to the size of nanoparticle. SEM images (Left) of the prismatic silver (a-c) and gold (d-f) nanoparticles monolayer arrays (in white color) made with nanosphere lithograph technique and the absorption spectra (Right) of the silver (a) and gold (b) periodic array samples made with the 0.26 (blue spectrum), 0.36 (green spectrum) and 0.45 μm (red spectrum) PS spheres, respectively. Expected dependence of surface plasmon absorption spectra on volume during the lattice oscillation of a gold or silver nanoparticle. Optically detected lattice phonon oscillations induced in the prismatic Ag (a,b,c) and Au (d,e,f) nanoparticles monolayer arrays with a 100 fs laser pulse at 400 nm and monitored near the absorption maximum of each nanoparticle (solid dots). The size (bisector) of the silver nanoparticles is around (a) 52.4, (b) 79.6 and (c) 99.3 nm. The size of the Au nanoparticles is around (d) 60.5, (e) 85.7 and (f) 103.7 nm. Wenyu Huang, Wei Qian and Mostafa A. El-Sayed, *Nano Letters*, 2004, 4 (9), 1741-1747; Wenyu Huang, Wei Qian and Mostafa A. El-Sayed, *The Journal of Physical Chemistry B*, 2005, 109 (40), 18881-18887.

C. Contributions to the field of Nano-Sensors:

Plasmonic nanoparticles are characterized by having the localized surface plasmon resonance (LSPR) spectrum and possessing strong surface electromagnetic plasmonic field. The peak position of LSPR spectrum of any plasmonic nanoparticles depends on the dielectric of the surrounding medium in addition to other physical parameters. The plasmonic field greatly enhances the Raman signal by a factor of over 10^{10} times. Both the shift of the LSPR peak position upon changing the dielectric function of the

surrounding medium and the enhancement of the Raman signal by the plasmonic field are used in the chemical and biological sensing technology. Recently, we have prepared plasmonic nanoparticles of different structures such as gold nanoframes¹ which when used in sensing showed very high efficiency². These gold nanoframes were used successfully to detect the Raman signal of toxic gases of low concentrations down to the zeptomolar level.¹ The unique efficiency of such nanoparticles is attributed to the presence of two types of plasmonic surfaces; one on the exterior surface of the particle and the other is on the interior surface of the cavity.³ The coupling between these two surface fields makes the LSPR peak position of these nanoparticles tunable in the visible and NIR regions.³ In order to make the plasmonic nanoparticles prepared by the colloidal chemical technique more applicable in sensing, we were able to assemble them into monolayer on the surface of substrate.⁴ The effect of substrate on the sensing efficiency of the plasmonic nanoparticles was discussed in our recent studies.^{5,6} The LSPR phenomenon is clearly sensitive to the nanoparticle shape of gold and silver nanostructure. We compared the sensing efficiency of nanoparticles made of gold and silver nanorods.⁷

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Contributions to the field of Medicine (Nanomedicine):

a. Introduction:

The most useful and important contributions of El-Sayed's group to nanoscience that best impacted Nanotechnology is indeed to the field of medicine and molecular cell biology. Their early results on the plasmonic properties of gold and silver nanoparticles^(1,2,3,4) enabled them to use what they found to develop methods for cancer diagnostic⁽⁵⁾ and photothermal therapy⁽⁶⁾. In order to use the property of gold nanoparticles in photothermal therapy in which moderately strong light is absorbed, converted into intense scattered light used in imaging applications or some of it is converted into high temperature heat used in the photo-thermal selective destruction of the cancer or any diseased cells. Only near infra-red light can penetrate long distances in the human body. Nanorods are known to absorb in the near IR at different wavelengths depending on the rods aspect ratios. We developed⁽⁷⁾ the best synthetic method to make the nanorods. This method is now the most used method for making the nanorod for use in photothermal and many other photonic applications Worldwide. This synthetic publication has won the most cited and most downloaded paper of the American Chemical society "Chemistry of material" journal every year for the past six years.

El-Sayed received five years funding from the National institute of Health (NIH) to use the gold nanorods in photo-thermal cancer therapy in animal studies jointly with the assistant director of the Winship Cancer Institute of Emory University Medical School in Atlanta. In addition, El-Sayed is directing a30-scientists research group at the Egyptian Research Center in Cairo, Egypt, funded by an Islamic group (Misr El-kheir) using the Zakah funds collected to help good causes for suffering people in Egypt, like those fighting cancer. These two studies are now in progress in Egypt and in the United States.

In another direction, the El-Sayed group has developed a spectroscopic technique making use of the strong imaging power of the enhanced scattering properties of gold or silver nanoparticles to image single cell behavior (using enhanced Rayleigh type scattering) or the molecular changes anywhere in a live single cell (using Surface Enhanced Raman Scattering, SERS and specific conjugation techniques). More important, these studies can be carried out in time during the cell lifetime, during its division or as the cell dies⁸ from any disease or as it is treated with any Cancer drug .

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b. Plasmonic Photo-Thermal Therapy (PPTT) of Cancer: Animal studies:

Owing to their characteristic Localized Surface Plasmon Resonance (LSPR), gold nanoparticles (AuNPs) which results from the absorption of light with resonant frequency and induce the coherent motion of the its conduction band electrons. This produces large surface electromagnetic fields. These decay either by giving off strongly enhanced light scattering (used for imaging) or via nonradiative processes (heat) resulting from the collisions between the excited oscillating electrons and the ionic lattice of the gold or silver metals (electron-phonon relaxation processes). We measured this to occur in the one picosecond timescale. We measured the time the hot metallic particle transfer its thermal energy to the surrounding (i.e. to the cancer cell to which it is attached) to be in the few hundreds of picoseconds resulting in its melting and destruction(1). The ability of these particles to heat the surrounding medium upon excitation, along with the sensitivity of cancer cells to hyperthermia, is exploited in the cancer treatment by the method we called the plasmonic photothermal therapy (PPTT) therapy. Plasmonic AuNPs are ideal candidates as photothermal contrast agents. They can be specifically attached to cancer cells via molecular targeting or systemically delivered to tumor sites via the enhanced permeability and retention (EPR) effect. Upon accumulation of the AuNPs at the disease site of interest, they can be triggered via external radiation, causing temperature increases characteristic of hyperthermia, and subsequently inducing apoptosis. Not only can the AuNPs target the disease site of interest for targeted PPTT, but their LSPR is tuned such that the particles can be triggered by near-infrared (NIR) light (650–900 nm) that easily passes through physiological tissues with minimal interference with water or the hemoglobin.

Our group demonstrated successful destruction of human oral squamous carcinoma cells (HSC-3), by selectively labelling them with antiEGFR conjugated AuNPs and irradiating them with a continuous wave (cw) visible laser to induce cell death (2). We later demonstrated the use of AuNRs for the NIR-triggered PPTT of HSC-3 tumor cells. Initially, we used antibody-conjugated AuNRs for the selective labeling of cancer cells in vitro.(3) Upon attachment of these photothermal contrast agents to the cell membrane,

near IR light (NIR) laser exposure activated the AuNRs, such that they released enough heat to induce membrane destruction in the labeled cancer cells leaving the healthy cells unaffected. After receiving the National institute of health funding in the USA, our group with the group of Dr Shin of Emory cancer center used this same technique into the in vivo regime, utilizing PEG-AuNRs that were not specific for cellular labeling, but were biocompatible and able to accumulate in HSC-3 tumor tissue due to the EPR effect.(4) These photothermal contrast agents were systemically delivered to mice, allowed to accumulate in tumor tissue, and then selectively activated via NIR laser irradiation, resulting in dramatic reduction in tumor growth. Recently, we demonstrated the optimization of AuNR-mediated PPTT (in vitro) by altering the AuNR's optical properties through nanoparticle size manipulation.(5). At the same time, two groups of Scientists at NRC in Cairo Egypt are doing studies on Mice and Rats and now moving to do it on Dogs. El-Sayed go to Egypt four times a year to give seminars and to direct the research at NRC and getting the NRC group in continuous contact with medical doctors at the National Cancer Center in Cairo. The minister of Health in Cairo is kept informed with the results of these groups and is kept informed by Professor El-Sayed with the progress in this field..

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c. Determination of Cancer Cell Death Mechanisms, drug efficacy and drug Delivery by the observed changes in time of single cell or of its molecular contents using plasmonic enhanced imaging:

The use of the plasmonic (strong scattering and photo-thermal) properties of gold and silver nanoparticles made Nanotechnology a naturally growing field due to its integration into many biomedical applications. Within nanotechnology, gold nanoparticles (AuNPs) are some of the most heavily utilized nanomaterials due to their unique optical, physical

and chemical properties.[1] In particular, their enhanced light scattering provides several advantages when used in biological imaging or as sensing agent. When compared to commonly used fluorophores, AuNPs' scattered light is 4-5 orders of magnitude brighter.

Our group has made use of the enhanced light scattering, to not only image cellular environments, but also to continually observe real-time intracellular molecular changes during various cellular processes. In 2010, AuNP scattering in conjunction with dark field imaging was used to determine particle localization at the nucleus and to monitor cellular response after a cytotoxic level of AuNPs were delivered.[2] This foundational work was expanded upon in 2012, as non-cytotoxic concentrations of nuclear-targeted AuNPs were used to probe the real-time molecular changes at the nucleus during cell cycle progression.[3] In particular, AuNP scattering was exploited to simultaneously acquire structural and molecular information from a single-cell through Rayleigh and Raman imaging, and revealed a strong correlation between DNA and protein Raman vibrations and cell cycle phase progression. Intracellular AuNP coupled Raman spectroscopy was also used to determine drug efficacy and drug delivery in cancer cells after treatment with chemotherapeutic agents.[4, 5] This work demonstrated the ability of AuNP light scattering to continually assess the efficiency of drug delivery and drug efficacy in real-time as well as reduced the time required to obtain cytotoxicity drug profiles.

Moreover, nuclear AuNP scattering and Raman spectroscopy were used to reveal real-time molecular changes in human cancer cells during apoptosis⁶. The ability to directly compare morphological changes (i.e. Rayleigh scattering images) with molecular changes (i.e. Raman spectroscopy analysis) of the DNA/protein composition around the cell nucleus revealed the occurrence and dynamics of three apoptotic molecular events: protein denaturation, proteolysis, and DNA fragmentation. A temporal profile of these molecular events was also created. Together, these results concluded apoptotic cell death mechanism and demonstrated the potential of AuNP scattering to advance molecular and cellular biology methods to accurately assess therapeutic imaging and response.

By loading large number of drug molecules on nanoparticle surfaces and deliver them to the sick cell it kills the cell instantaneously minimizing the possibility of drug resistance to occur. The only question raised when this does not happen is whether or not the drug was actually delivered by being released from the surface of the nanoparticle. Due to their unique optical and physical properties,⁷ plasmonic nanoparticles, specifically of gold nanoparticles have been heavily utilized in biomedical applications. Recently, we have employed gold nanoparticles to not only deliver the chemotherapeutic drug doxorubicin, but to also monitor its delivery in real-time using a new technique termed plasmonically-tunable Raman/fluorescence imaging spectroscopy (P-TRFIS).⁸ This technique exploits the ability of gold nanoparticles to selectively turn "ON" or "OFF" Raman and fluorescence signals based upon the drug molecules distance from the nanoparticle surface. Doxorubicin (DOX), a common chemotherapeutic drug, was covalently attached to the gold nanoparticle surface using a pH-sensitive hydrozone linkage that is susceptible to cleavage in acidic environments, such as present in the cell

lysosomes. When the DOX-gold nanoparticle conjugate is initially administered to cancer cells, the Raman signals of DOX are greatly enhanced, while its fluorescence is quenched due to its proximity to the plasmonic field of the nanoparticles. Once the drug-nanoparticle conjugated entered the lysosome, the DOX is released from the nanoparticle resulting in the restoration of its fluorescence intensity and the disappearance of its Raman spectra. as it moves away from the thenanoparticle plasmoic field. By simultaneously tracking the intensity of the fluorescence and Raman spectra of DOX functionalized gold nanoparticles, we were able to monitor the drug delivery in real-time on the single cell level !

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d. Summary of the Nano-medicine research at Georgia Tech :

Gold nanoparticles exhibit a combination of physical, chemical, optical, and electronic properties unique from other biomedical nanotechnologies and provide a highly multifunctional platform with which to image and diagnose diseases, to selectively deliver therapeutic agents, to sensitize cells and tissues to treatment regimens, to monitor and guide surgical procedures, and to preferentially administer electromagnetic radiation to disease sites^{1, 2}. Because of their large size, circulating nanoparticles preferentially accumulate at tumor sites and in inflamed tissues due to the characteristically defective architecture of the vessels which supply oxygen and nutrients to these tissues. Once circulating nanoparticles extravasate through these large vascular pores and into the disease site, they remain lodged due characteristically diminished lymphatic drainage and their low diffusivity. Gold nanoparticles can be used to deliver drugs or imaging contrast agents that otherwise exhibit poor solubility or pharmacokinetic profiles, or to deliver agents which are intrinsically susceptible to enzymatic degradation and/or exhibit poor intracellular penetration such as small interfering RNA (siRNA). Nanogold can be routinely surface functionalized with active ligands at densities that are 100- and 1000-fold higher than that achievable with conventional liposomes or PLGA nanoparticles, respectively, allowing their binding affinity to be optimized for a particular disease, type, stage, or patient. Because of their comparability in size to the distances between cell-surface targets, gold nanostructures can simultaneously engage multiple, adjacent receptor sites, achieving increased selectivity in their uptake through this multivalent avidity. The novel optical and electronic properties of gold nanoparticles are also particularly attractive for use in multimodal drug delivery applications where these structures can afford enhanced drug pharmacokinetics/biodistribution and simultaneous hyperthermia and radiation therapy contrast, as well as photo-imaging contrast, spectro-chemical diagnostic contrast, and, when molecularly directed to specific sub-cellular sites, intrinsic pharmaco-dynamic properties.

. **Thus Biomedical Research in El-Sayed Lab** is aimed to leverage the unique structural, optical, and electronic properties of gold nanotechnologies to improve cancer diagnostics and therapy and to understand molecular cell biology. These strategies broadly emphasize the development of nanoscale tools for **(i)** molecularly-targeted cancer diagnostics, **(ii)** laser-photo-thermal tumor therapy, and **(iii)** understanding the fundamental interactions of nanoscale materials with biological systems.

Early studies in his laboratory asked whether the unique optical properties of gold nanoparticles could be used to more easily discriminate cells based on the presence or absence of cellular biomarkers. Using antibody-gold nanoparticle conjugates targeting the cell-surface receptor, EGFR, they found that malignant cells could be selectively labeled with gold nanoparticles in vitro, providing high optical scattering contrast for identification using a simple light microscope or optical spectrometer.³ In later work they extended this approach to develop multivalent analogues of small molecule inhibitors that block cancer cell signaling.⁴ They found multivalent display improved drug performance several fold per drug molecule and could be extended to a variety of other therapeutic cell signaling cascades.

Another aspect of his research asks whether targeted gold nanoparticle delivery can facilitate improved cancer diagnostics and therapeutics in vivo. By designing rod-shaped gold nanoparticles capable of absorbing tissue-penetrating light waves and converting them into heat, they were able to show, for the first time, that gold nanorods could serve as efficient and systemically administrable contrast agents for laser photothermal tumor therapy.⁵ They observed 2-4 fold increased heating in nanoparticle-treated tumors, resulting in 50% complete remission in a mouse cohort receiving their optimized treatment protocol. Extending this approach, they have also studied how electromagnetic enhancement by gold nanoparticles can augment cell killing in photodynamic therapy, where photoexcitation of small molecule 'photosensitizers' leads to the formation of toxic oxygen species. By tailoring gold nanoparticle size/shape to facilitate resonant electronic interaction with the photosensitizer, they were able to improve cell killing by up to 200-fold while greatly decreasing dose-limiting toxic effects.

In parallel, their research also sought to better understand the ways that nanoscale materials can interact with and manipulate biological systems. By molecularly targeting gold or silver nanoparticles to discrete sub-cellular locations, they found that they can not only selectively perturb various biological process,⁶ but also monitor in real-time⁷ their dynamics via using rationally-designed nanoparticles that enhance, for example, Raman scattering or fluorescence quenching⁸.

To summarize, El-Sayed's group biomedical research interfaces biomolecular recognition with the unique optical and electromagnetic properties of nanoscale noble metals, in particular, gold due to its nontoxic properties. These investigations seek to improve cancer detection and drug potency, while also developing new methods for remediating solid tumors and manipulating biological systems via nanoscale technologies. In addition, El-Sayed is now supervising two groups, one in the states and supported by NIH and another one in Egypt supported by the people of Egypt. In this work, the positive results his group obtained in the in-vitro cell experiments are being extended to cancer-ridden animals e.g. mice, rats and now dogs.

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3. El-Sayed, I.H., Huang, X. & El-Sayed, M.A. Selective laser photo-thermal therapy of epithelial carcinoma using anti-EGFR antibody conjugated gold nanoparticles. *Cancer Lett.* **239**, 129-135 (2006).
4. Dreaden, E.C., Mwakwari, S.C., Sodji, Q.H., Oyelere, A.K. & El-Sayed, M.A. Tamoxifen-Poly (ethylene glycol)-Thiol Gold Nanoparticle Conjugates: Enhanced Potency and Selective Delivery for Breast Cancer Treatment. *Bioconjugate Chem.* **20**, 2247-2253 (2009).
5. Dickerson, E.B. et al. Gold nanorod assisted near-infrared plasmonic photothermal therapy (PPTT) of squamous cell carcinoma in mice. *Cancer Lett.* **269**, 57-66 (2008).
6. Kang, B., Mackey, M.A. & El-Sayed, M.A. Nuclear Targeting of Gold Nanoparticles

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7. Kang, B., Austin, L.A. & El-Sayed, M.A. Real-Time Molecular Imaging throughout the Entire Cell Cycle by Targeted Plasmonic-Enhanced Rayleigh/Raman Spectroscopy. *Nano Lett.* **12**, 5369-5375 (2012).
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Research Support

The research activity of El-Sayed's group has been supported in the past 50 years by ~18 M dollars, mostly from the Department of Energy, the National Science Foundation, Office of Naval Research, USA Air force and NIH. At GeorgiaTech, he has continuous funding from the Julius Brown chair. The research funding he had has contributed to the education of over 100 Ph.D. students, 12 M.Sc.students and over 45 postdoctoral fellows. His laboratory attracted over 30 visiting Professors and scholars supported by their own funding. By mid of 2014, Over 670 papers were published, mostly in primary journal that received over .

Since joining Georgia Tech in 1994, El-Sayed supervised the research of over 25 graduate students, ten undergraduate students (four RUE students) and over seventeen postdoctoral fellows. His laboratory has attracted National and international Professors to spend time and carry out research in his Laboratory.

At Georgia Tech, he and his group published over 340 papers in twenty years, mostly in peer reviewed journals of high impact factors. In the last 5 years the number of citations/paper has exceeded the one hundred

Research Training (Research Collaborators)

Postdoctoral or Research Scientist Fellows

Dr. T. Pavlopoulos (1962-1964) (Navy Research Lab, San Diego)
Dr. K. Eisenthal (1964-1965) (Columbia University)
Dr. J. K. Roy (1963-1965)
Dr. N. K. Chaudhuri (1964-1967) (University of Calcutta)
Dr. M. Bhaumik (1966) (Northrop)
Dr. W. Moomaw (1965-1966)(Tufts University)
Dr. D. Tinti (1967-1970)(University of California, Davis)

Dr. Eva Migirdicyan (1968-1969) (University of Paris)
Dr. T. Kuan (1969-1971) (Northrop)
Dr. O. Kalman (1969-1970)
Dr. A. Shain (1970-1971) (DuPont)
Dr. P. Esherick (1973-1975) (Sandia)
Dr. P. Zinsli (1973-1975) (University of Bern)
Dr. S. Sheng (1974-1976)
Dr. P. Avouris (1975-1977) (IBM)
Dr. J. Berg (1976-1978) (TRW)
Dr. R. Moncorges (1977)
Dr. A. Burns (1978-1980) (Sandia)
Dr. Jai-Hyung Lee (1980-1982) (Seoul National University)
Dr. J. Simon (1983-1984) (University of California, San Diego)
Dr. P. Dupuis (1983-1985) (University of Montreal, Canada)
Dr. P. Evesque (1984-1985) (CNRS, Paris)
Dr. K. Ismail (1985-1986) (University of Qatar)
Dr. E. Chronister (1985-1987) (University of California, Irvine)
Dr. A. Eychmuller (1987-1988) (Hahn Meitner Institute)
Dr. D.-J. Jang (1987-1989) (Seoul National University)
Dr. R. van den Berg (1988-1990) (Shell, Amsterdam)
Dr. Kuang-Jung (Gloria) Hsu (1989) (National Taiwan University, Taiwan)
Dr Stephen Lukanov (1993-1997) (Moscow State University, now a research Scientist at Coening))

Dr. Kyo Yoo Jung (1994-95) (Han Yang University, South Korea)
Dr. Valey Kamalov (1995-1996) (Chemical Phys. Institute, Russian Academy of Science)
Dr. Bingsou Zou (1997-1999)
Dr. Clemens Burda (1997-1998) Now a professor at Case Western Reserve
Dr. Jianping Wang (1999-)
Dr. Markus Braun (2000-2001) (University of Stuttgart)
Dr. Stephan Link (2000-200, now a Professor at Rice U.)
Dr. Dr. Wei Qian (2002-Present) (Research Scientist II)
Dr. Kyeong-Seok Lee (2005-2006, now head of a division at KIST, South Korea)
Dr. Xiaohua Huang (2006-2009, now Professor at U of Memphis)
Dr. Sujatha Pushpanketh (2006-2007) (Research Scientist)
Dr. Nasrin Hooshmand (2007 and 2013-Present, she is a Professor in Iran)
Dr. Mahmoud Mahmoud (2007-Present)
Dr. Svetlana Neretina (2007-2009, now Professor at Temple University)
Dr Li Song
Dr N. Allam (Professor at the American U Of Cairo)
Dr F Saira (now a Professor in Pakistan)
Dr H. Nishikion (Professor in Japan)
Dr Bin Kang (Professor at the Technicat University in Nanjing, China)
Dr Paul Szymanski (Assistant to the director, LDL)
Dr X Kang
Dr Megan Mackey (now a Research Scientist at Merk & Co)
Dr Sajanlal
Dr Steve Hira

Visiting Professors and Scholars

Prof. John Olmsted, American University of Beirut (1970-1971)

Prof. Peter Wagner, Michigan State (1971-1972)
 Prof. W. Moomaw, Williams College (1971-1972)
 Professor W. R. Ware, University of Western Ontario (1974-1975)
 Anne-Marie Merle, University of Bordeaux, (1977-1978)
 Prof. J. Prochorow, Polish Academy of Sciences (September - April 1979)
 Dr. B. Karvaly, Hungarian Academy of Sciences (March 1979 - August 1981)
 Dr. Francisco Pessine, University of Campinas, Brazil January 15 - March 15, 1980
 Professor Paras Prasad - Spring 1981
 Dr. Baohua Guo, Lanzhou Institute of Chemical Physics, Visiting Scholar (1982-1984)
 Dr. G. Comtet, CNRS Fellow (1983-1984)
 Dr. H. Niederwald, Deutsch Forschungsgemeinschaft Fellow (1983-1984)
 Prof. E. Awad, American University of Beirut (1985-1987)
 Prof. H. Morita, Chiba University, Japan (1989-1990)
 Prof. E. Awad, American University of Beirut (1988-1994)
 Prof. F. W. Froben, Freie University of Berlin (1991)
 Prof. Ch. Bräuchle, University of Munich (1992) and (1993)
Prof. Kyung-Hoon Jung, WonKwang University, Korea (1994-1995)
Dr. V. Kamalov, Chemical Physics Institute, Russian Academy of Sciences (1995-1996)
Prof. A. Henglein, Hans Meitner Institute (Spring, 1995)
Prof. M. Abdel Kader, Cairo University (1998)
Dr. Akihiro Furube, Osaka University (Fall 1999)
Prof. Yusei Miyagi, University of the Ryukyus, Okinawa, Japan (2000)
Prof. Curtis Shannon, Auburn University (2001)
Dr. Marie-Paule Pileni, Paris, France (2003)
Prof. Ramazan Öztürk" Turkey
Sir John Thomas, United Kingdom (2007)
Prof. Hassan Talaat, Cairo, Egypt (2007)
Professor Nasrin Hooshmand, Iran, (2013-2014)
 Professor Xu, China.(2013-2014)
 Professor Bin Kang.China, 2012-2013

Ph.D. Students

Thesis Title

- | | |
|---------------|---|
| 1. A. Yencha | “Absorption and Ionization Studies in the Vacuum Ultraviolet,”
1968 |
| 2. B. Ziegler | “Heavy-Atom, Vibronic and Host-Crystal Perturbations on the
Phosphorescence of Molecular Crystals” 1968 |
| 3. D. Demeo | “Photoionization and Photodissociation Cross-Sections of Crystals,”
1969 |
| 4. E. Yee | “Triplet Energy Annihilation and Multiplication in Molecular Crystals,”
1969 |
| 5. L. Hall | “Spin-Lattice Relaxation and Zeeman Effects in the Triplet State of
Pyrazine,” 1971 |
| 6. D. Owens | “Effects of Magnetic Field, Microwaves, Temperature
Changes on Phosphorescence-in Mixed Molecular Crystals |
| 7. R. Chen | “Microwave Double Resonance Spectroscopy of 1,2,4,5,
Tetrachlorobenzene,” 1972 |
| 8. A.Gwaiz | “PMDR Spectroscopy of Aromatic Hydrocarbons and |

- their N-heterocyclics. 1972
9. C. Lin "Low Temperature Triplet Energy Traps in Organic Solids Using Phosphorescence-Microwave Double Resonance Spectroscopy: Monomer and Excimer," 1974
 10. J. Chodak "The Use of Phosphorescence Microwave Double Resonance in the Study of Static and Dynamic Molecular Properties," 1974
 11. A. Wilkerson "The Study of Inter- and Intramolecular Energy Transfer in Organic Molecular Crystals," 1974
 12. M. Leung "Spin-Selectivity of Triplet State Photochemical Reactions by the Techniques of Phosphorescence Microwave Double Resonance," 1974
 13. E. Gossett, Jr. "Polarized Microwave-Phosphorescence Double Resonance Spectroscopy," 1974
 14. R. Leyerle "Low-Field Zeeman Phosphorescence-Microwave Double-Resonance Spectroscopy and the Mechanism of Intersystem Crossing," 1974
 15. M. Souto "Properties of the Triplet States of Ketones and Cyclic Hydrocarbons by Phosphorescence-Microwave Double-Resonance Methods," 1974
 16. T. Akasheh "Perturbation Effects on Molecular and Resonance Pair Triplet Excited States," 1977
 17. A. Campion "Energy Transfer in Inorganic, Organic and Biological Materials: Time-Resolved Laser Spectroscopy Studies," 1977
 18. A. Merle "The Origin of the Multiple Sites in a Shpol'skii Matrix," 1978
 19. W. Pitts "Triplet-Spin Labels in Structural and Dynamic Studies of Mixed Aromatic Solids," 1978
 20. J. Turner "Time Resolved Resonance Raman Spectroscopy of Bacteriorhodopsin" 1979
 21. D. Parker "Multiphoton Ionization in Polyatomic Molecules," 1979
 22. W. Hopewell "Energy Transfer in Ionic and Molecular Systems," 1980
 23. C. L. Hsieh "Resonance Raman Spectroscopy of Bacteriorhodopsin: Photoisomerization of Retinal in Picosecond and Nanosecond Time Scales and Hydration Effects of Purple Membrane on Film," 1981

24. J. Lurie "Multiphoton Ionization-Dissociation of Nitrogen-Containing Compounds by Visible and Ultraviolet Lasers," 1981
25. J. Fukumoto "Protein and Chromophore Spectroscopy and Dynamics During the Photochemical Cycle of Bacteriorhodopsin, 1982.
26. R. Pandolfi "Multiphoton Ionization Mass Spectrometry by Visible and Ultraviolet Pulsed Lasers," 1982
27. J. Morgan "Dephasing and Energy Transfer in Disordered Solids," 1983
28. D. Gobeli "Dynamics of Laser Multiphoton Ionization Dissociation of Two, Four-Hexadiyne," 1984
29. Jeng-jiun Yang "Experimental and Theoretical Studies on the Mechanism of Laser Multiphoton Ionization-Fragmentation of Some Benzene Derivatives," 1984
30. R. St. Pierre "Reactions of Niobium Metal Clusters Toward Benzene and Other Cyclic Hydrocarbons," 1987
31. Tsong-Lin Tai "Linear Reflectron Time-of-Flight Mass Spectrometric Studies of Laser Ionic Multiphoton Dissociation Dynamics," 1987
32. Chasn-Ion Yang "Transport Processes in Disordered Systems: Energy Transfer, Electrodeposition and Simulation of Chemical Reactions on Surfaces," 1987
33. Tim Corcoran "The Role of Metal Cations in the Function of Bacteriorhodopsin," 1987
34. Du-Jeon Jang "Picosecond Spectroscopic Studies of Energy and Proton Transfers," 1987
35. Diane M. Szaflarski "Photodissociation Dynamics of Gas Phase Polyatomic Molecules Studied Using Laser Multiphoton Ionization Mass Spectrometry," 1988
36. Li Song "Size Dependence of the Reactivity of Gaseous Niobium and Vanadium Clusters," 1989
37. Hyun Jin Hwang "Translational Spectroscopy Studies on the Photo Dissociation of Organic Iodides and the Unimolecular Dissociation of $\text{Cs}(\text{CsI})_n^+$ Clusters," 1991
38. Laura Sweetman "The Effects of Various Perturbations on the Function of Bacteriorhodopsin," 1992

39. Shuguang Wu "Bacteriorhodopsin; Some Studies on its Structure, Function and Potential Use as Biomaterial," 1992
40. Gloria Lin "Studies of Perturbed and Mutated Purple Membranes Using Steady State and Time-Resolved Spectroscopies," 1992
41. John Freitas : "The Structure and Dissociation Dynamics of Some Organic Iodides by State Selective Photofragment Translational Spectroscopy," 1993.
42. Nancy Zhang "Studies of Cation Binding to Bacteriorhodopsin and the Effects of Certain Mutations on its Structure and Function" 1993.
43. Keith Fagerquist "The Effect of AgI Polar Molecules on the Dissociation Dynamics of Ag Clusters Cations and Gaseous Clusters" 1994.
44. Jennifer Griffiths "*Determination of the Location and Binding Mechanism of Metal Cations in Bacteriorhodopsin*" 1996
45. DiFei Yang "*Spectroscopic and Analytical Studies of Wild Type and Structurally Perturbed Bacteriorhodopsin*" 1996
46. Temer Ahmadi "*Colloidal Metallic and Gaseous Ionic Nanoparticles: Structure, Dynamics, and Catalysis*" 1996
47. Tina Masciangioli "*Structural and Dynamic Studies of Bacteriorhodopsin and its Variants*" 1999
48. Reginald Little "*Synthesis and Characterization of II-VI Semiconductor Quantum Dots, Quantum Shells, and Quantum Wells*" 1999
49. Jianping Wang "*Time-Resolved Spectroscopic Studies of the Dynamic Processes in Bacteriorhodopsin*" 1999
50. Travis Christopher Green "*Photo-Induced Charge Carrier Dynamics and Self-Organization in Semiconductor and Metallic Nanocrystals: In Between the Bulk and Individual Molecules*" 1999
51. Stephan Link "*Spectral Properties and Relaxation Dynamics of Surface Plasmon Electronic Oscillations in Gold and Silver Nanodots and Nanorods*" 2000
52. Victor Volkov "*Bacteriorhodopsin Excited State Dynamics and Photochemistry*" 2000
53. Janet Petroski "*Platinum Metal Nanoparticles: Investigation of the Shape, Surface, Catalysis, and Assembly*" 2001

54. Nikoobakht, B. *"Synthesis, characterization and self-assembly of gold nanorods and surface-enhanced Raman studies, Georgia Institute of Technology"* 2003
55. Colin Heyes *"Protein Stability in Bacteriorhodopsin"* 2003
56. Christy Landes *"The Molecular-Nanoparticle Semiconductor Transition"* 2003
57. David Hathcock *"Dynamic and Catalytic Properties of Some Metallic Nanocrystals"* 2004
58. Radha Narananyan *"Shape-Dependent Nanocatalysis and The Effect of Catalysis on The Shape and Size of Colloidal Metal Nanoparticles"* 2005
59. Laurie Sanii *"Bacteriorhodopsin: Structure/Function Relationship"* 2005
60. Xiaohua Huang *"Gold Nanoparticles Used in Cancer Cell Diagnostics, Selective Photothermal Therapy and Catalysis of NADH Oxidation Reaction"* 2006
61. Susan Eustis *"Gold and Silver Nanoparticles: Characterization of their Interesting Optical Properties and the Mechanism of their Photochemical Formation"* 2006
62. Alexander Schill *"Interesting Electronic and Dynamic Properties of Quantum Dot Quantum Wells & other Semiconductor Nanocrystal Heterostructures"* 2006
63. Qusai Darugar *"Surface Effects on the Ultrafast Electronic Relaxation of Some Semiconductor and Metallic Nanoparticles"* 2006
62. Alexander Schill *"Interesting Electronic and Dynamic Properties of Quantum Dot Quantum Wells & other Semiconductor Nanocrystal Heterostructures"* 2006
63. Qusai Darugar *"Surface Effects on the Ultrafast Electronic Relaxation of Some Semiconductor and Metallic Nanoparticles"* 2006
64. Wenyu Huang *"Fundamental studies of the interaction between femtosecond laser and patterned monolayer plasmonic nanostructures"* 2007
65. Prashant Jain *"Plasmons in assembled metal nanostructures: Radiative and non-radiative properties, near-field coupling and its universal scaling behavior"* 2008
66. Biesso, Arianna *"Plasmonic field effects on the spectroscopic and photobiological function of the photosynthetic system of bacteriorhodopsin"* 2008
67. Christopher Tabor *"Some optical and catalytic properties of metal nanoparticles"* 2009

68. Chun-Wan Yen *“Plasmon photochemistry on the nanoscale” 2011*
69. Erik Dreaden *“Chemistry, photophysics, and biomedical applications of gold nanotechnologies” 2012*
70. Steven Hayden *“Novel applications of nanotechnology in medicine and green energy” 2012*
71. Rachel Near *“Theoretical and experimental investigation of the plasmonic properties of noble metal nanoparticles” 2013*
72. Megan Mackey *“Gold Nanoparticles in Chemical and Photothermal Applications of Cancer Therapy” 2013*
73. Lauren Austin *“Exploring Some Aspects of Cancer Cell Biology with Plasmonic Nanoparticles” 2014*

M. Sc. Students

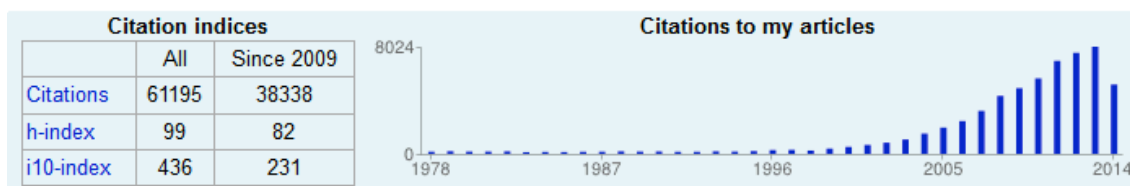
- S. Wilt “Vibration Spectra and Structure of 1,4,7-Nonatriene,” June, 1964 (M.Sc.)
- R. Lam “Phosphorescence Microwave Double Resonance Studies of Triphenylamine Molecules,” September, 1974 (M.Sc)
- Y. Morita “The Effect of Ultraviolet Radiation and Oxygen on the Light Adaptation Process of Bacteriorhodopsin,” October, 1978 (M.Sc)
- L. Hashimoto “Investigation of the Origin of the Exciton CD Spectrum of Bacteriorhodopsin,” January , 1983 (M.Sc)
- J.Hanamoto “Investigation of the Role of Tyrosine in the Proton Pump Cycle of Bacteriorhodopsin,” Winter 1984 (M.Sc.)
- E. Jackson “The Determination of the Rotational and Vibrational Distributions of .OH Formed from the Dissociation of (H₂O)₂ through Laser Induced Fluorescence” June 1990, (M.Sc); September, 1990 (C.Phil.)
- Adam Poncheri *Plasmonic field effects of silver nanoparticle monolayers on poly(phenylene ethynylene) fluorescent polymers of different chain length” 2011*
- Snyder, Brian *“An investigation into bimetallic hollow nanoparticles in catalysis”*

Publications:

List of Published Research - M. A. El-Sayed

Publications from 1 to 332 are on research carried out at UCLA in the field of chemical Physics, spectroscopy and electronic energy relaxation in molecules and matter in different phases while publications from 333 present are on research carried out at LDL (Georgia Tech): <http://ldl.gatech.edu> started in 1994 and mostly in the field of Nanoscience and Nanotechnology. The relative citations to this work with time is given in the graph below and by titles and numbers, please go to: <http://scholar.google.com/citations?user=BMPDLscAAAAI&hl=en>

List of Publications:



**PART I
ELECTRONIC
RELAXATION
UCLA
1962-1994**

**PART II
NANOSCIENCE &
NANOTECHNOLOGY
GEORGIA TECH
1995-**

PART I: Research on Molecular Spectroscopy & electronic Relaxation: (Carried out at the University of California: 1962-1996)

1. M.A.El-Sayed and R. K. Sheline, "The Infrared Spectrum and Structure of Hexacyanodnickelate (I) Ion, $[\text{Ni}_2(\text{CN})_6]^{-4}$," J. Amer. Chem. Soc., 78 (1956).
2. M.A.El-Sayed and R. Wolfgang, "Chemical Reaction of Recoil Tritium with Gaseous Alkanes," J. Amer. Chem. Soc. 79, 3286 (1957).
3. M.A.El-Sayed, Peter Estrup, and R. Wolfgang, "Mechanism of the Reaction of Recoil Hydrogen in the Gaseous Phase," J. Phys. Chem., 62, 1356 (1958).
4. M.A.El-Sayed, " A Nuclear Method for Determining Very Low Vapor Pressures," Nucl. Instru., 3, 359 (1958).
5. M.A.El-Sayed and R. K. Sheline, "The Infrared Spectrum and Structure of the $[\text{Ni}(\text{CN})_4]^{-4}$ Ion," J. Amer. Chem. Soc., 80, 2047 (1958).
6. M.A.El-Sayed and R.K. Sheline, "The Position of the CN Stretching Frequency in Organic and Inorganic Molecules," J. Inorg. Nucl.Chem., 6, 187 (1958).

7. M.A.El-Sayed, "The Relation Between the CN Stretching Frequency and Hammett's ρ ," *J. Inorg. Nucl.Chem.*, 10, 168 (1959).
8. M.A.El-Sayed and M. Kasha, "Orbital Type Interchange by Solvation and Effects on the Emission Properties of Naphthalene N-heterocyclics," *Spectrochimica Acta*, 15, 758 (1960).
9. M.A.El-Sayed, M. Kasha, and Y. Tanaka, "Ionization Potentials of Benzene, Hexadeuterobenzene, and Pyridine from their Observed Rydberg Series in the 600-2000 Å," *J. Chem. Phys.*, 34, 334 (1961).
10. M.A.El-Sayed and G. W. Robinson, "Excitation Transfer Splitting in the n, π^* Transitions of the Diazines," *J. Chem. Phys.*, 34, 1840 (1961).
11. M.A.El-Sayed and G. W. Robinson, "Comments on a Communication by El-Sayed and Robinson," *J. Chem. Phys.*, 35, 1896 (1961).
12. M.A.El-Sayed and G. W. Robinson, "Intramolecular Excitation Transfer. The Lowest n, π^* Transitions in Pyrazine," *Mol. Phys.*, 4, 273 (1961).
13. M.A.El-Sayed, "The Effect of Reducing the Symmetry on the Spectra of Benzene in the 1500-2000 Å Region: Spectra of Pyridine, Pyrimidine, and Pyrazine," *J. Chem. Phys.*, 36, 552 (1962).
14. M.A.El-Sayed, "The Radiationless Processes Involving Change of Multiplicity in the Diazenes," *J. Chem. Phys.*, 36, 573 (1962).
15. M.A.El-Sayed, "Perturbational Enhancement of the Coupling Between the Lowest Two Electronic p, p^* States in Naphthalene," *J. Chem. Phys.*, 36, 1943 (1962).
16. M.A.El-Sayed and H. D. Kaesz, "Assignment of the CO Stretching Absorptions in C_4v Metal Pentacarbonyl Derivatives," *J. Mol. Spect.*, 9, 310 (1962).
17. M.A.El-Sayed, M. T. Wauk, and G. W. Robinson, "Retardation of Singlet and Triplet Excitation Migration in Organic Crystals by Isotopic Dilution," *Mol. Phys.*, 5, 205 (1962).
18. W. Rhodes and M. A. El-Sayed, "Observed Electronic Transitions in Hexahelicene," *J. Mol. Spect.*, 9, 42 (1962).
19. M.A.El-Sayed, "Intramolecular Resonance Interaction Between Fundamental Vibrations of Polyatomic Molecules," *J. Chem. Phys.*, 37, 680 (1962).
20. G. Dodson, M. A. El-Sayed, I. Stolz, and R. K. Sheline, "Photochemical Formation of Some Metal Hexacarbonylacetonitrile Derivatives," *Inorg.Chem.*, 1, 526 (1962).
21. M.A.El-Sayed, "The Method of Oscillating Interacting Dipoles and the Vibrational Spectra of Some Organic and Metal Poly-Carbonyl Derivatives," *Spectrochimica Acta*, 18, 1387 (1962).
22. M.A.El-Sayed, "S-T Radiationless Process and the Emission Properties of Nitrogen Heterocyclics," *Bull. Amer. Phys. Soc.*, 7, 499 (1962).
23. M.A.El-Sayed, "Proposed Effect of High Pressures on the Radiationless Processes," *J. Chem. Phys.*, 37, 1568 (1962).
24. M.A.El-Sayed and H. D. Kaesz, "Infrared Spectra and Structure of the Tetracarbonyl Halide Dimers of Manganese, Technetium and Rhenium," *Inorg.Chem.*, 2, 158 (1963).
25. M.A.El-Sayed, "Origin of the Phosphorescence Radiation in Aromatic Hydrocarbons," *Nature*, 197, 481 (1963).
26. M.A.El-Sayed, "Comments on Contaminating the Ground State with Triplet Character," *J. Chem. Phys.*, 38, 3032 (1963).
27. M.A.El-Sayed, "Spin Orbit Coupling and the Radiationless Processes in Nitrogen Heterocyclics," *J. Chem. Phys.*, 38, 2834 (1963).
28. M.A.El-Sayed, "Polarization of Molecular Luminescence in Plastic Media by the Method of Photoselection," *J. Opt. Soc. Amer.*, 53, 797 (1963).

29. M.A.El-Sayed and T. Pavlopoulos, "Polarization of the Triplet-Triplet Absorption Spectrum of Some Polyacenes by the Method of Photoselection," *J. Chem. Phys.*, 39, 834 (1963).
30. M.A.El-Sayed and R. G. Brewer, "Polarization of δ , δ^* and n , δ^* Phosphorescence Spectra of N-heterocyclics," *J. Chem. Phys.*, 39, 162 (1963).
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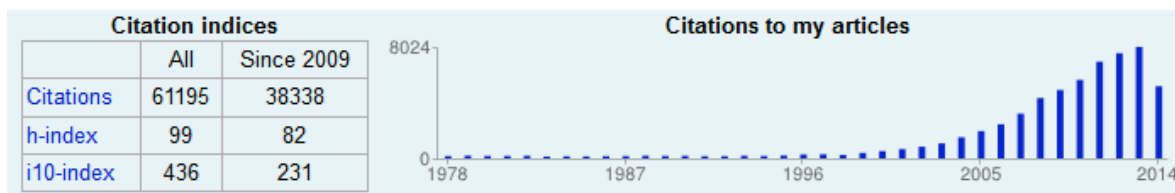
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