

# Notes in GC-MS

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# Outline

GC-MS system

Gas phase chromatography

Gas, filters and injection

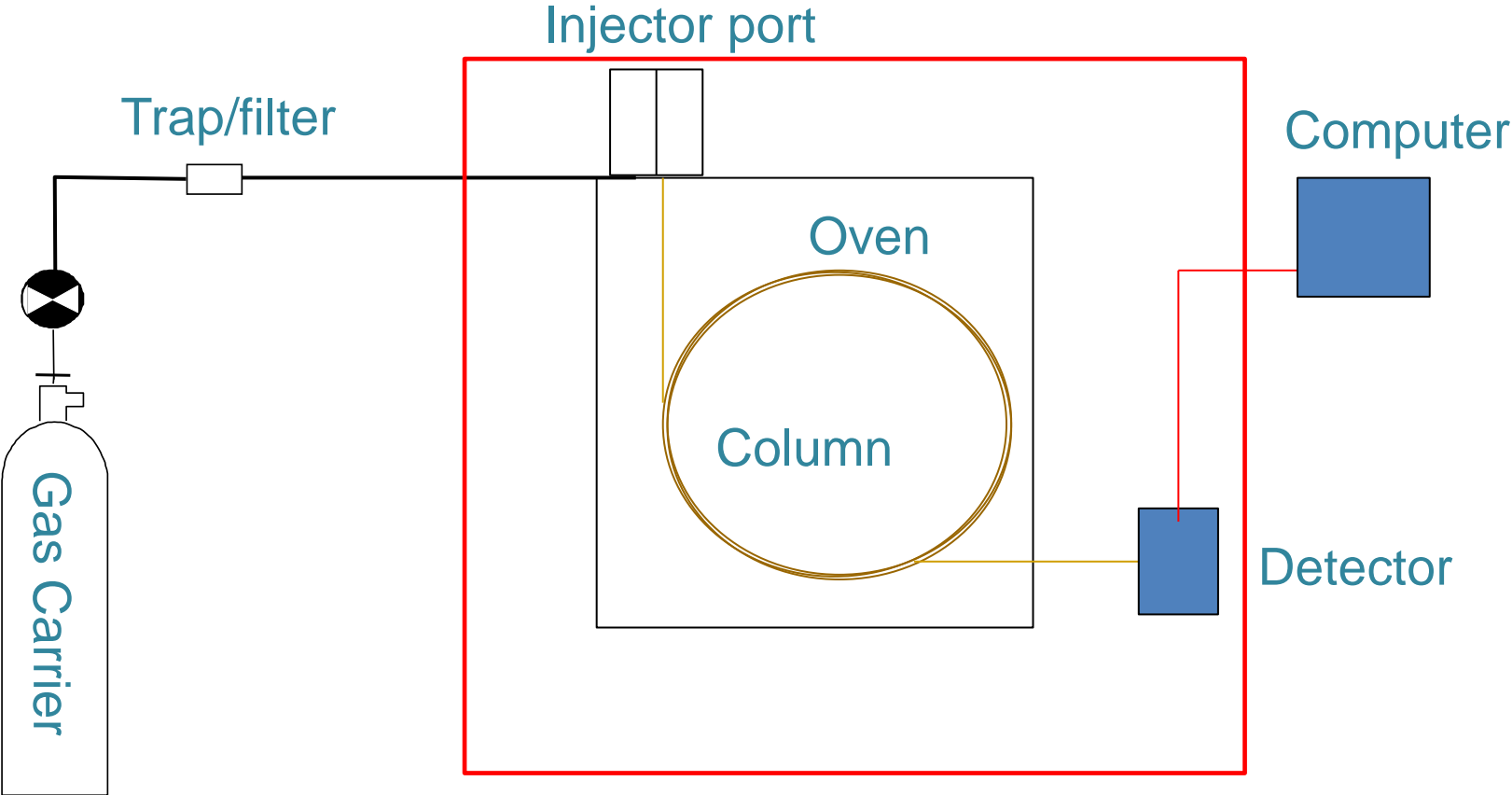
Injector. Split/splitless

The column

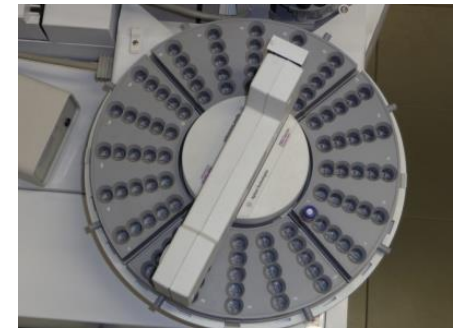
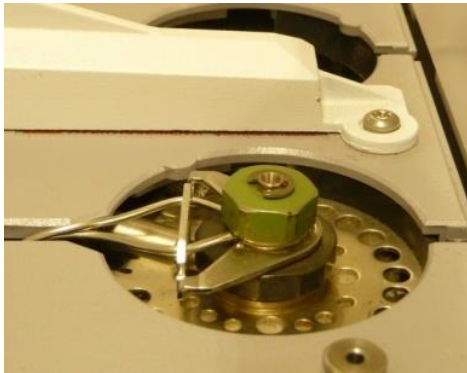
Detectors. Mass detector

Quantitation

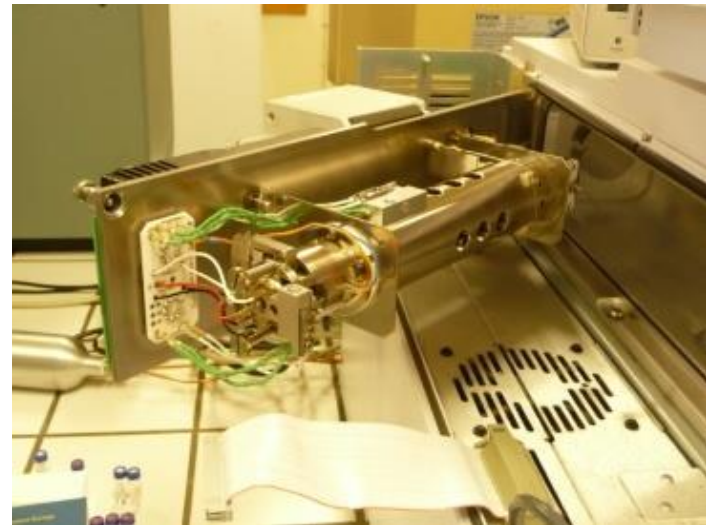
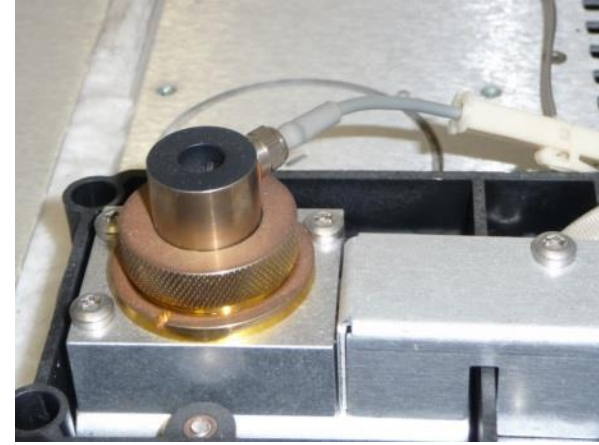
# Schematic Diagram of a Gas Chromatograph



# Gas carrier tanks, gas filters and injection



# Oven and detector mass detector



# Gas phase chromatography

The mobile phase is a gas

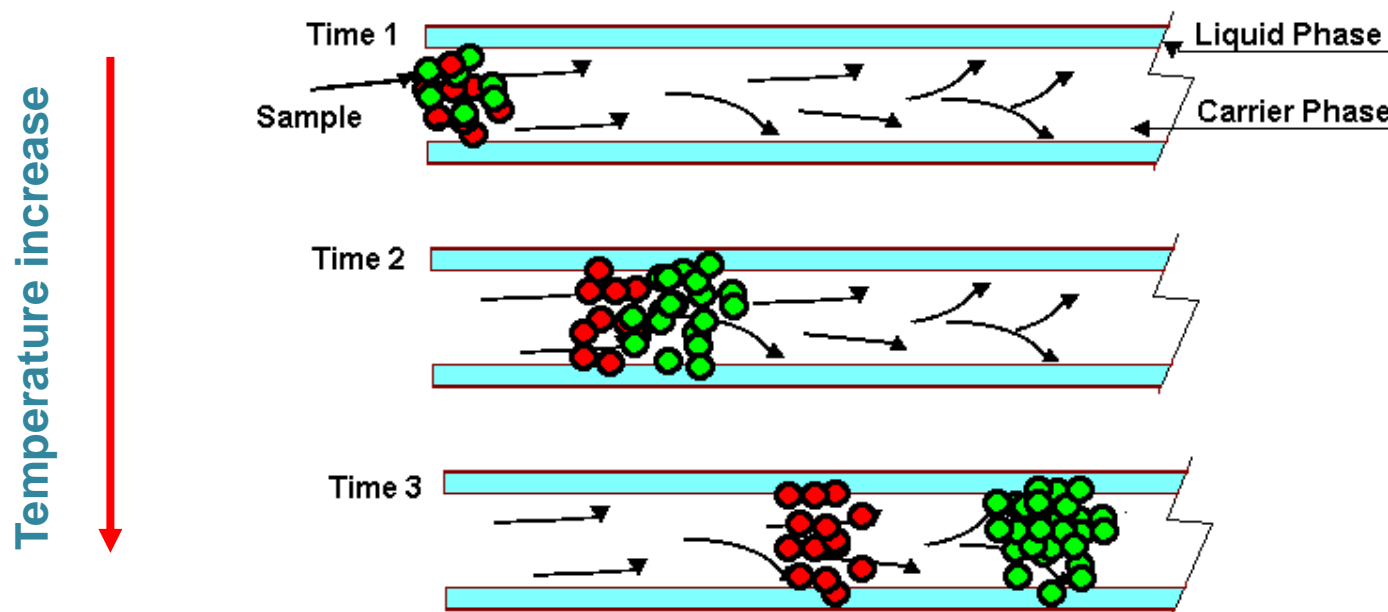
The stationary phase is a liquid or a solid. Partition/adsorption

The sample must be vaporized and injected onto the head of the chromatographic column.

The sample is transported through the column by the mobile phase flow.

The differential interaction of the vapor phase analytes with the stationary phase gives rise to the chromatographic separation

# The separation process



Which compounds elute first?

Most volatile / lowest boiling point normally elutes first

# Advantages and disadvantages of Gas Chromatography

**a) Fast analysis**

**b) High Resolution**

**c) Sensitive detectors**

(easy ppm, often ppb)

**d) Highly accurate quantification**

**e) Automated systems**

**f) Non-destructive**

In the sense that it allows online coupling to Mass Spec.

**g) Small sample ( $\mu\text{L}$ )**

**h) Reliable and relatively simple**

**i) Low cost**

**a) Limited to volatile samples**

Do not inject samples with non volatile compounds – DMSO, salts, proteins, etc

**b) Not suitable for thermally labile compounds**

**c) Some samples may require extensive preparation**

**d) Requires spectroscopy (usually MS) to confirm peak identify**



# The gas

For high-speed analysis the ratio of viscosity to diffusion coefficient should be as small as possible. **H<sub>2</sub>** would be the best choice, followed by **helium**.

Hydrogen is the fastest carrier gas ( $u_{opt}$ ), with an optimum linear velocity of 40 cm/sec, and exhibits the flattest van Deemter profile. **Hydrogen is explosive when concentrations exceed 4% in air.**

Helium is the next best choice, with an optimum linear velocity of  $u_{opt} = 20$  cm/sec. **Expensive**

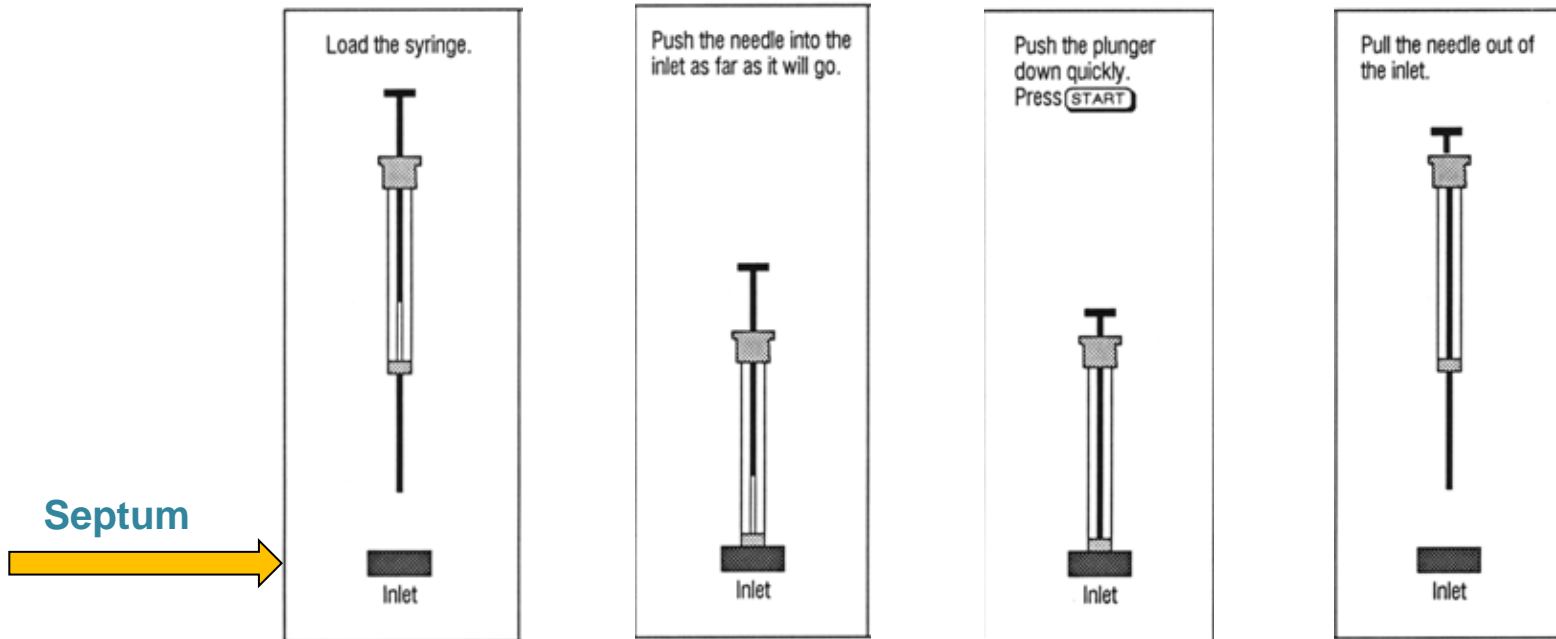
The **purity** of the carrier should be at least **99.995%** for best results.

The most common impurities are **air**, **water** and **hydrocarbons**.

Impurities can lead to sample decomposition, column deterioration, detector contamination and deterioration, a change the response factors, a change in the detection limits.

# Manual injection

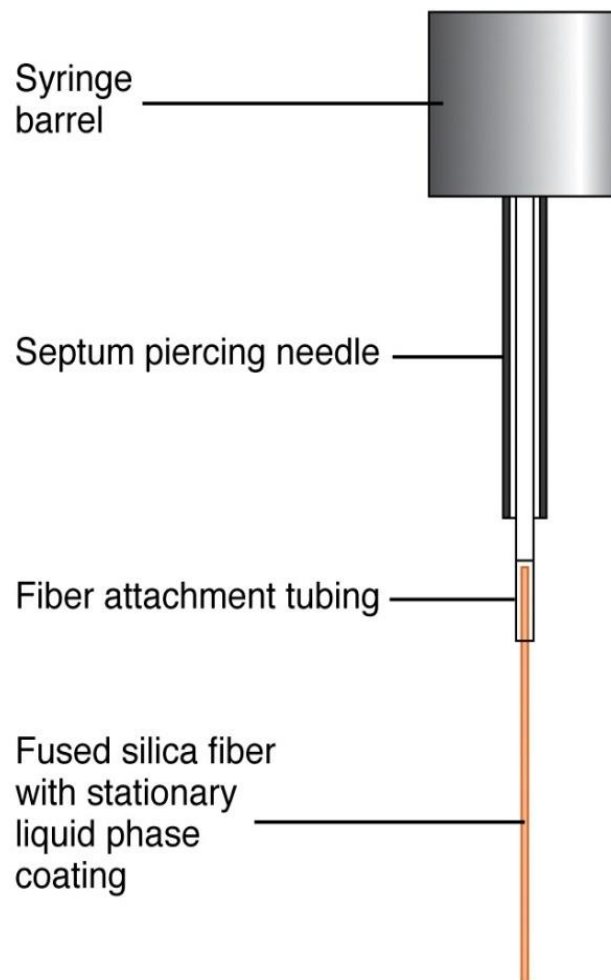
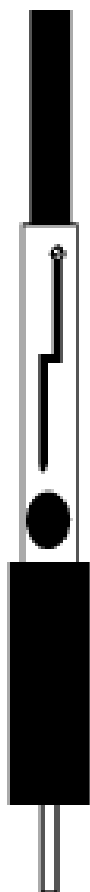
Typically 1.0  $\mu\text{L}$  or less using a microsyringe. No air bubbles



- 1 - The injection should be made in a single quick motion.
- 2 - The whole operation should only take 3 to 4 seconds.

# HS-Solid phase micro-extraction (HS-SPME)

Syringe for solid phase microextraction.



# HS-Solid phase micro-extraction (HS-SPME)

## Extraction Procedure

Pierce septum on  
sample container.



Expose SPME  
fiber/extract analytes.



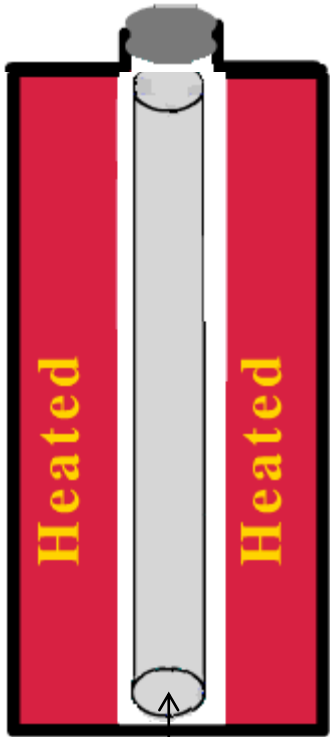
Retract fiber/withdraw  
needle.



Expose the fiber to a solution or headspace for fixed time under stirring.

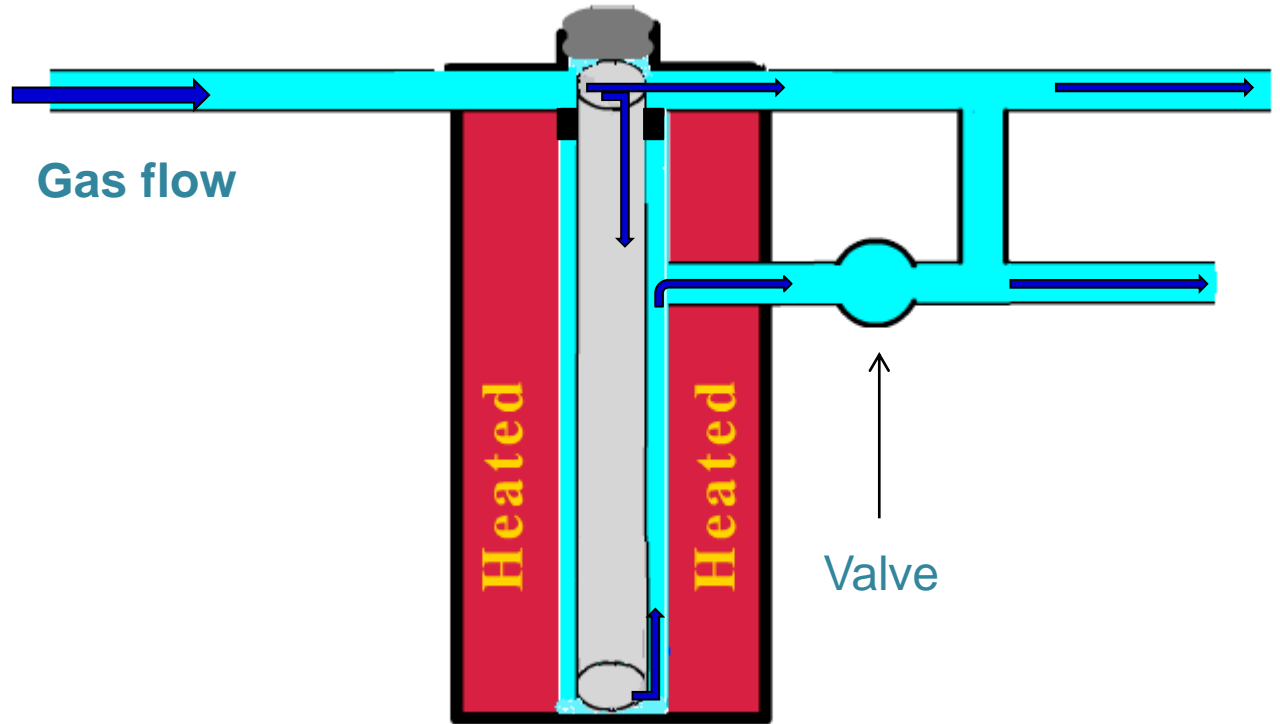
# The injector

Septum



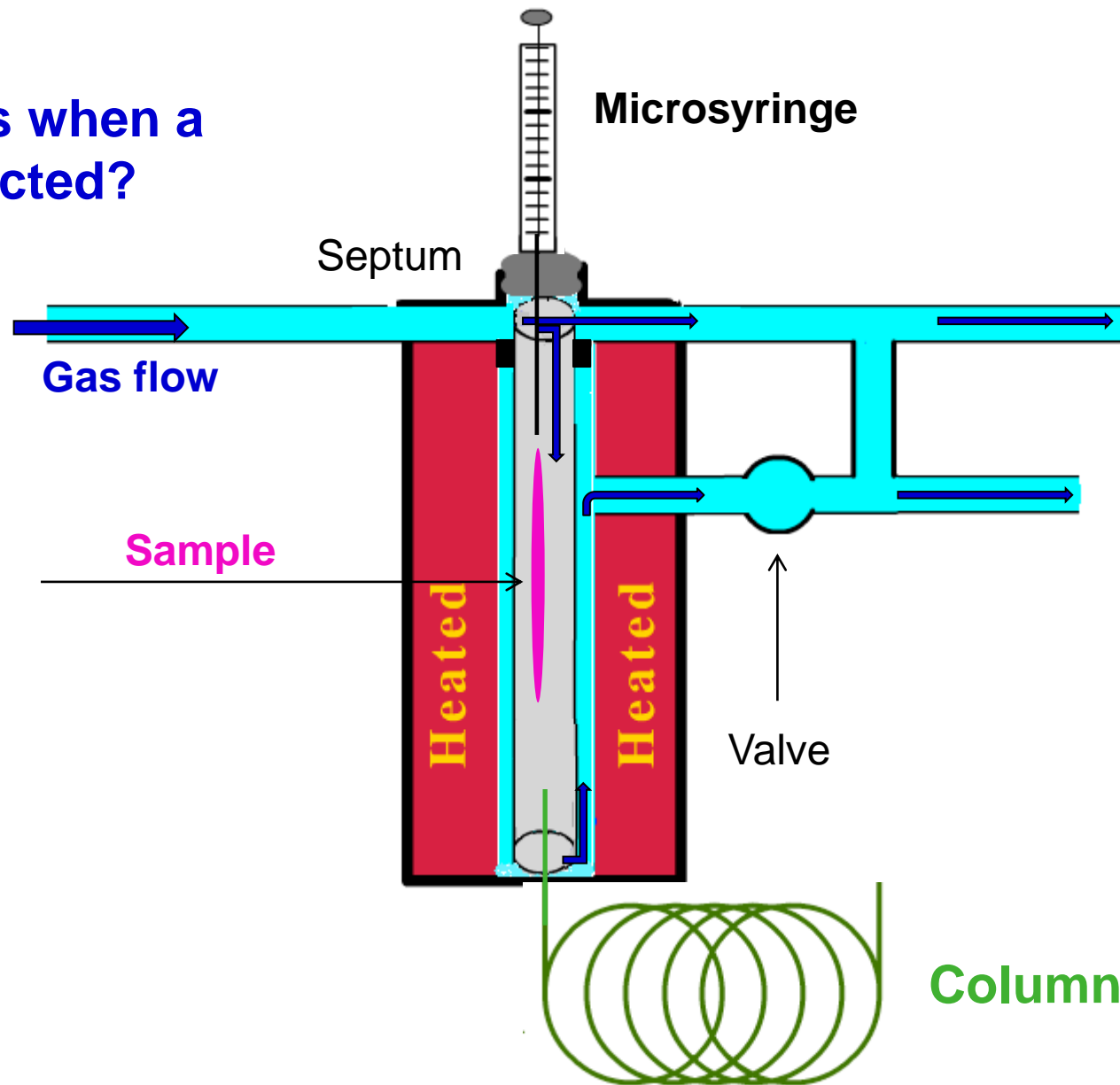
Liner

# The injector and the flow system

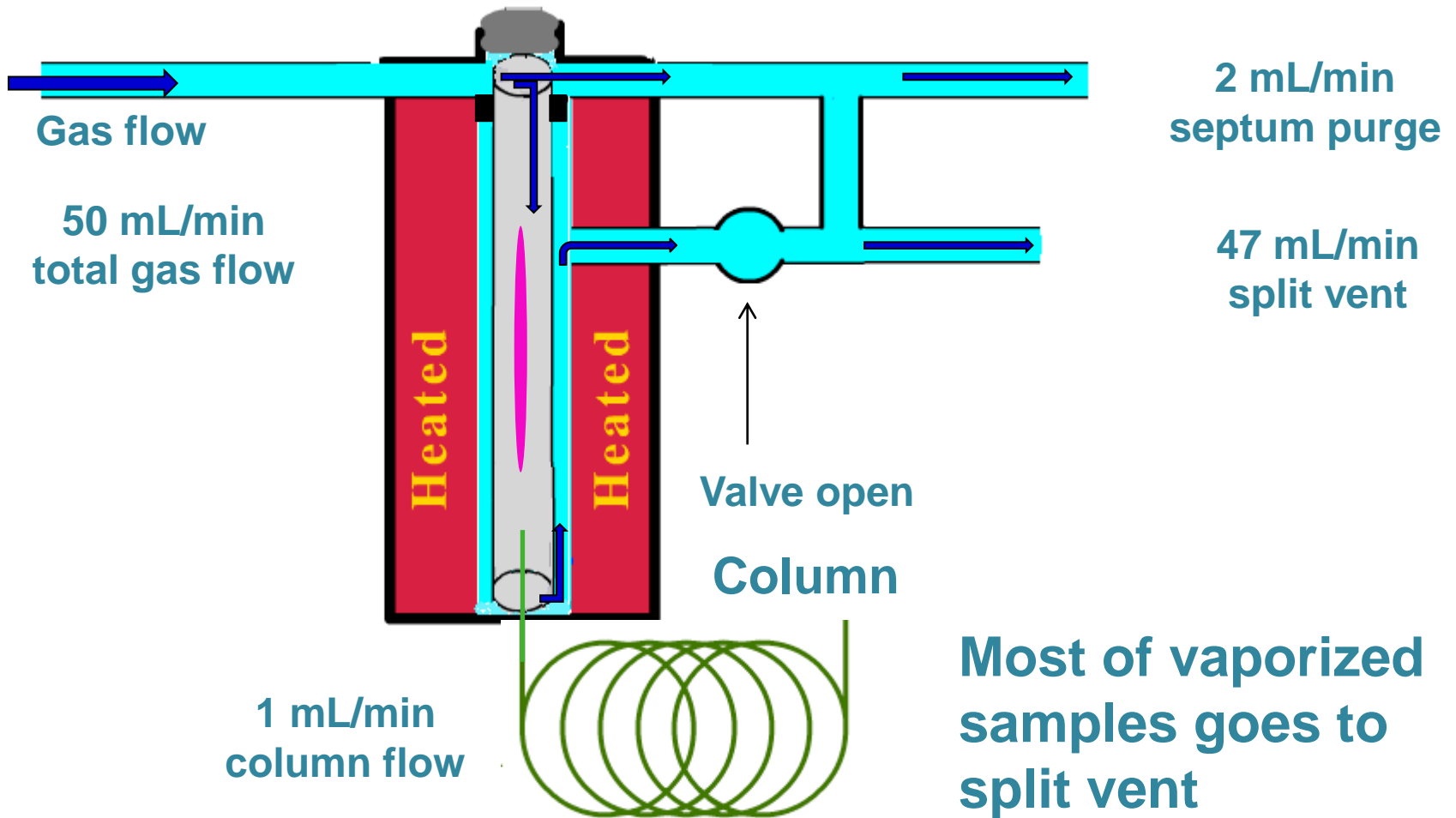


Valve

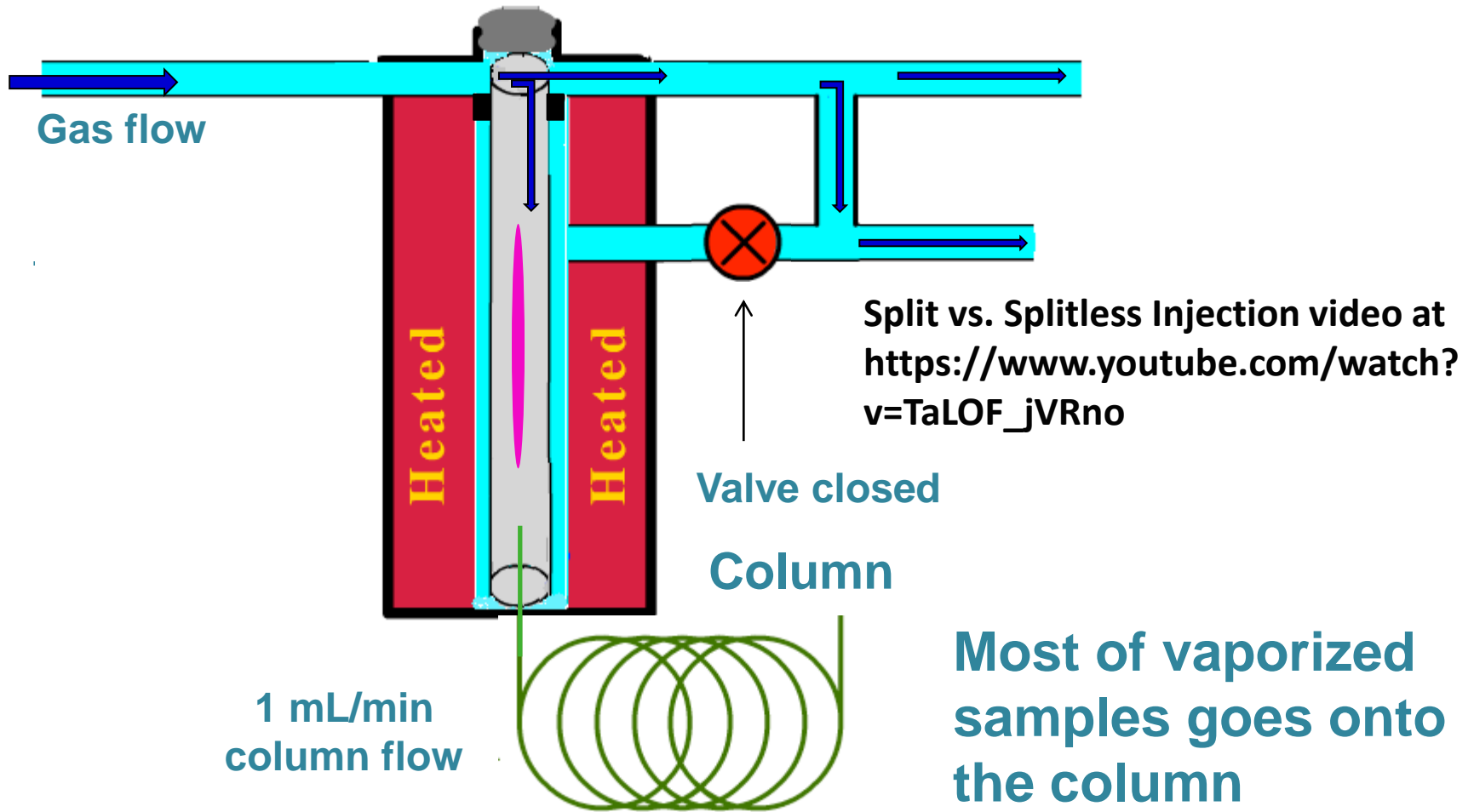
What happens when a sample is injected?



# The split mode



# The splitless mode





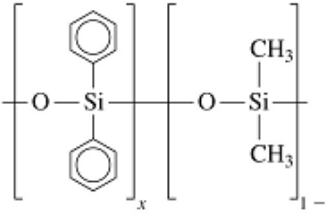
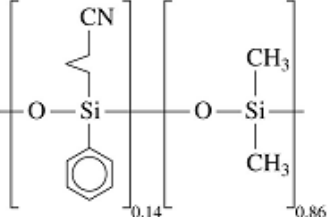
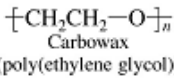
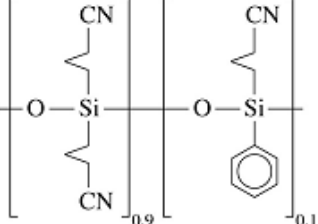
# Open tubular columns

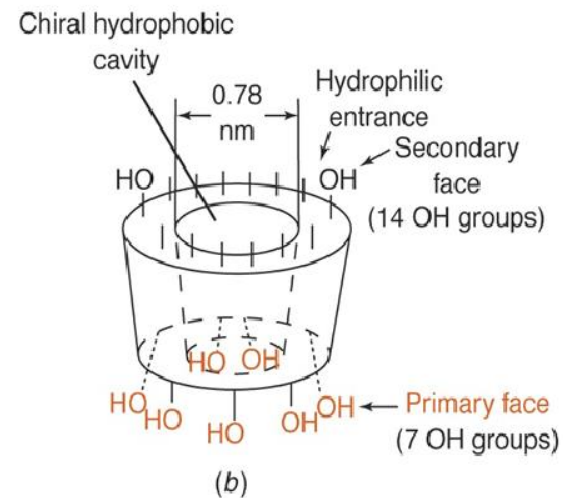
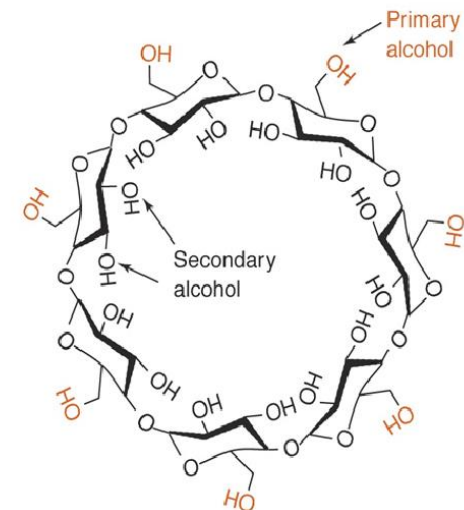
Check the column properties before use



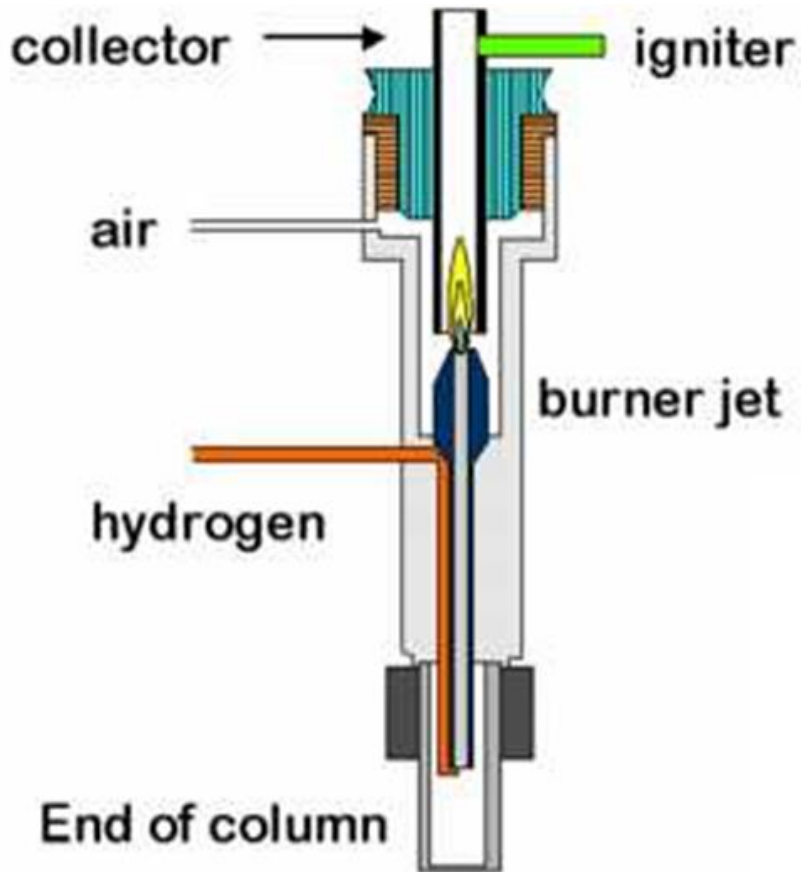
	Thin-film narrow-bore	Thick-film narrow-bore	Thick-film wide-bore
Description	Thin-film narrow-bore	Thick-film narrow-bore	Thick-film wide-bore
Inner diameter	0.10–0.32 mm	0.25–0.32 mm	0.53 mm
Film thickness	~0.2 $\mu\text{m}$	~1–2 $\mu\text{m}$	~2–5 $\mu\text{m}$
Advantages	High resolution Trace analysis Fast separations Low temperatures Elute high b.p. compounds	Good capacity Good resolution (4 000 plates/m) Easy to use Retains volatile compounds Good for mass spectrometry	High capacity (100 ng/ solute) Good for thermal conductivity and infrared detectors Simple injection techniques
Disadvantages	Low capacity ( $\leq 1$ ng per solute) Requires high sensitivity detector (not mass spectrometry) Surface activity of exposed silica	Moderate resolution Long retention time for high b.p. compounds	Low resolution (500–2 000 plates/m) Long retention time for high b.p. compounds

### Common stationary phases in capillary gas chromatography

Structure	Polarity	Temperature range (°C)
 <p>(Diphenyl)<sub>x</sub>(dimethyl)<sub>1-x</sub> polysiloxane</p>	$x = 0$ Nonpolar	-60°–320°
	$x = 0.05$ Nonpolar	-60°–320°
	$x = 0.35$ Intermediate polarity	0°–300°
	$x = 0.65$ Intermediate polarity	50°–370°
 <p>(Cyanopropylphenyl)<sub>0.14</sub>(dimethyl)<sub>0.86</sub> polysiloxane</p>	Intermediate polarity	-20°–280°
 <p>Carbowax (poly(ethylene glycol))</p>	Strongly polar	40°–250°
 <p>(Biscyanopropyl)<sub>0.9</sub>(cyanopropylphenyl)<sub>0.1</sub> polysiloxane</p>	Strongly polar	0°–275°



# Flame Ionization Detector



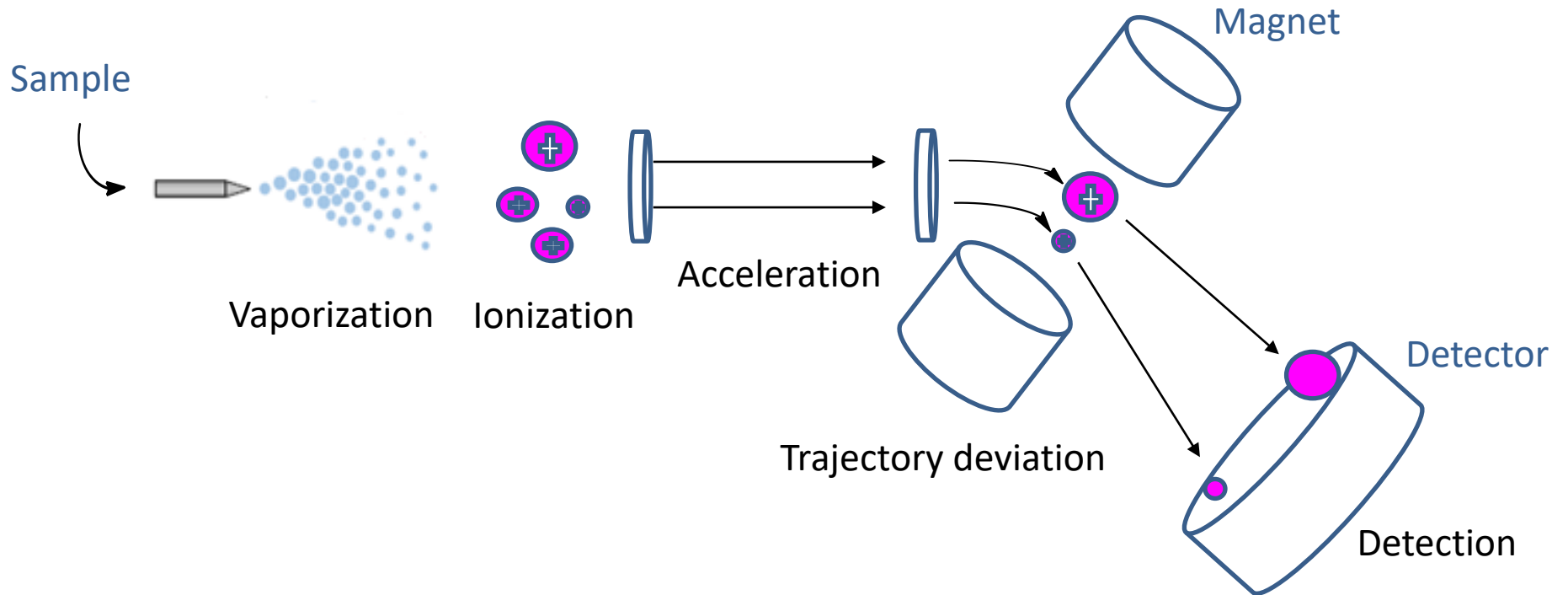
- 1 - sample burned in H<sub>2</sub>/air
- 2 - flame
- 3 - sample must be combustible
- 4 - must use electrometer ppm sensitivity
- 5 - destructive

# What is mass spectrometry?

Mass spectrometry is the “weighing” of individual molecules by transforming them into ions in vacuo and then measuring the response of their trajectories to electric and magnetic fields or both.

John B. Fenn et al., Science, 246, 64, 1989

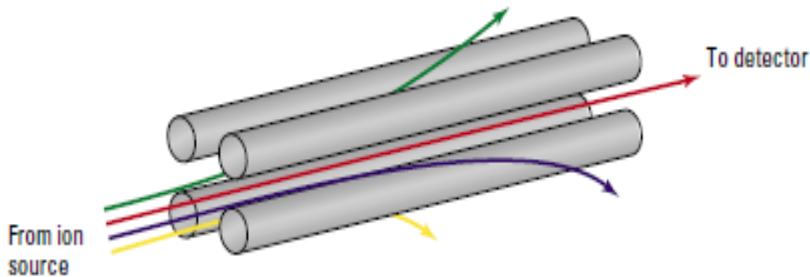
# Mass spectrometer, how does it work?



# Mass analysis

## Quadrupole

a) A quadrupole mass analyzer consists of four parallel rods arranged in a square.



b) Voltages applied to the rods generate electromagnetic fields.

d) Quadrupoles tend to be the simplest and least expensive mass analyzers.

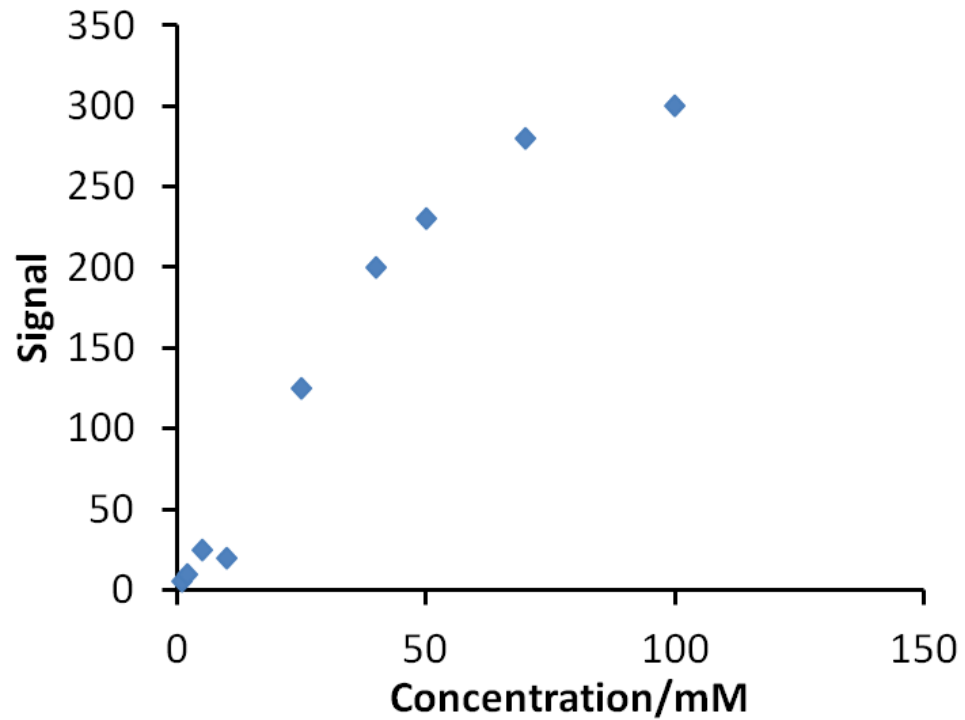
e) Quadrupole mass analyzers can operate in two modes:

- Scanning (scan) mode
- Selected ion monitoring (SIM) mode – quantitative, target compounds

# Quantitative analysis

- 1 - It usually involves the use of the peak areas (sometimes the peak height)
- 2 – Needs the comparison of the area (or height) of analyte peak with that of one or more standards.
- 3 – Calibration curve

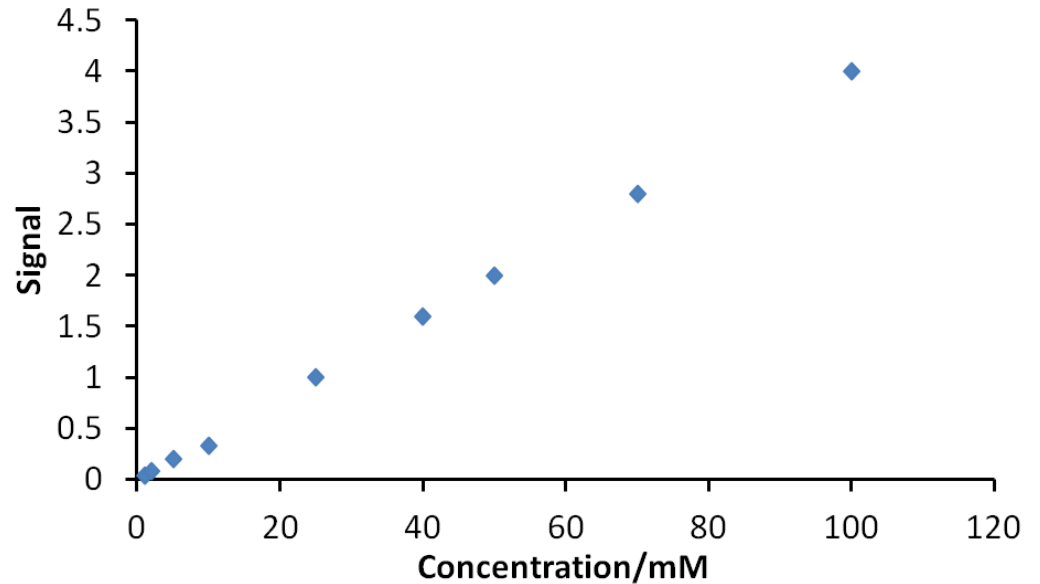
Conc	Signal
1	5
2	10
5	25
10	20
25	125
40	200
50	230
70	280
100	300



# Quantitative analysis

## The quantification using an internal standard

Conc	Signal	Signal	IS
1	0.04	5	125
2	0.08	10	125
5	0.2	25	125
10	0.333	20	60
25	1	125	125
40	1.6	200	125
50	2	230	115
70	2.8	280	100
100	4	300	75



**Its use corrected the uncertainties introduced by the sample injection**



## Remarks

**1 – Do not inject solution with particles**

**2– Volatile and semi-volatile compounds only**

No salts, such as NaCl, borates, phosphates or other

**3 – Only compounds stable in the gas phase**

**4 – The injection of samples containing non-volatile compounds such as salts and high molecular weight compounds will contaminate the liner and might block the column**

**5 – Do not inject water**

**6 – For GC-MS use a solvent delay (for example 2 min) to preserve the filament**

## Selected videos

### **Replacing Your Liner, Septum and O-Ring**

<https://www.youtube.com/watch?v=mExb2mj0eEk>

### **Replacing the Gold Seal**

<https://www.youtube.com/watch?v=uMTKMbmSvRI>

### **GC Column Installation - Part 1**

<https://www.youtube.com/watch?v=Nx1FzgwICiE>

### **GC Column Installation - Part 2**

<https://www.youtube.com/watch?v=YmkKikaakSI>